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Development of three-dimensional printing polymer-ceramic scaffolds with enhanced compressive properties and tuneable resorption

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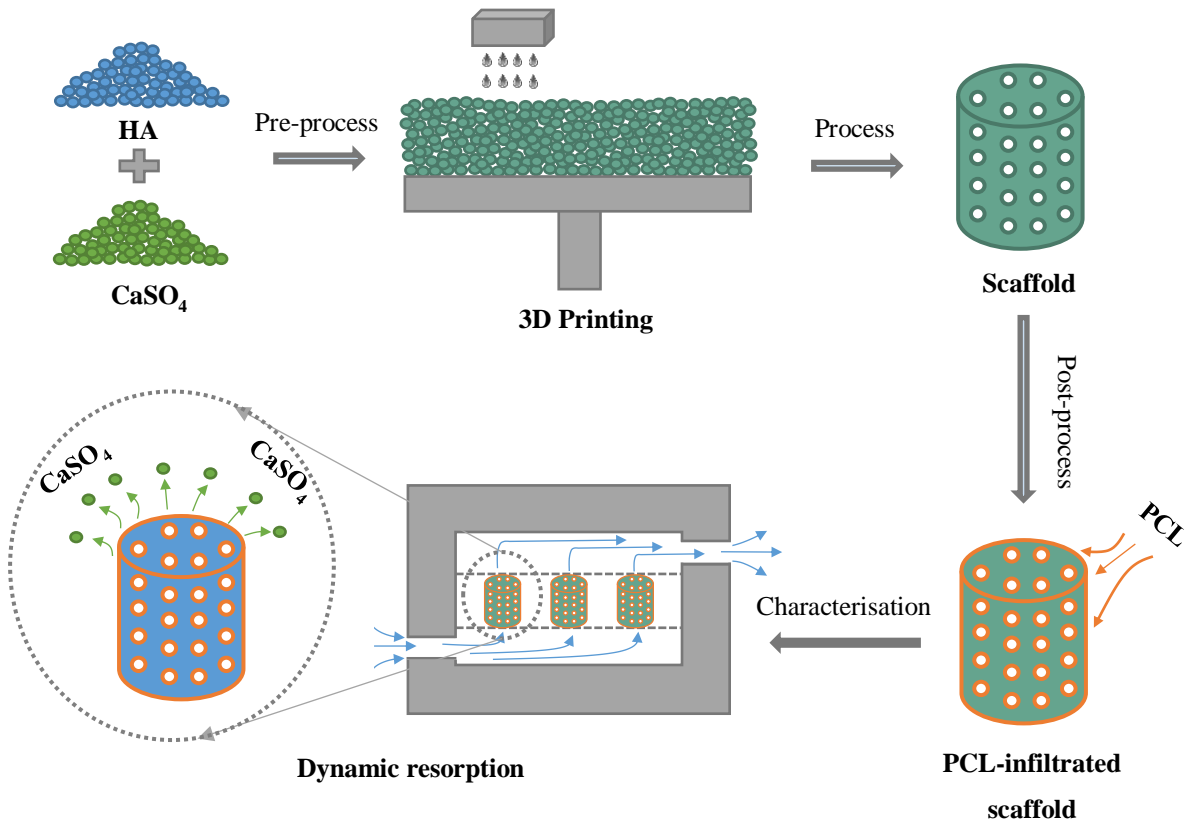
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Graphical Abstract



Highlights

- HA:CaSO₄ 3DP scaffolds represent a promising prospect for tuning bioresorption.
- PCL infiltration of the 3DP scaffold significantly improved compressive properties.
- *In vitro* resorption under dynamic flow demonstrated PCL inhibited HA dissolution.
- CaSO₄ resorption created niches within the scaffold, encouraging new tissue growth.
- PCL provided long-term structural support irrespective of *in vitro* resorption.

1 **Development of Three-Dimensional Printing Polymer-Ceramic Scaffolds with Enhanced**
2 **Compressive Properties and Tuneable Resorption**

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29 **Abstract**

30 In this study, bone tissue engineered scaffolds fabricated via powder-based 3D printing from
31 hydroxyapatite (HA) and calcium sulphate (CaSO₄) powders were investigated. The
32 combination of using a fast resorbing CaSO₄ based powder and the relatively slower HA
33 powder represents a promising prospect for tuning the bioresorption of 3D printed (3DP)
34 scaffolds. These properties could then be tailored to coincide with tissue growth rate for
35 different surgical procedures. The manufactured scaffolds were infiltrated with poly(ε-
36 caprolactone) (PCL). The PCL infiltrated the inter-particle spacing within the 3DP structures
37 due to the nature of a loosely-packed powder bed and also covered the surface of ceramic-
38 based scaffolds. Consequently, the average compressive strength, compressive modulus and
39 toughness increased by 314%, 465% and 867%, respectively. The resorption behaviour of the
40 3DP scaffolds was characterised *in vitro* using a high-throughput system that mimicked the
41 physiological environment and dynamic flow conditions relevant to the human body. A rapid
42 release of CaSO₄ between Day 0 and 28 was commensurate with a reduction in scaffold mass
43 and compressive properties, as well as an increase in medium absorption. In spite of this, HA
44 particles, connected by PCL fibrils, remained within the microstructure after 56 days resorption
45 under dynamic conditions. Consequently, a high level of structural integrity was maintained
46 within the 3DP scaffold. This study presented a porous PCL-HA-CaSO₄ 3DP structure with the
47 potential to encourage new tissue growth during the initial stages of implantation and also
48 offering sufficient structural and mechanical support during the bone healing phase.

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51 **Keywords**

52 Additive Manufacturing; 3D printing; hydroxyapatite; calcium sulphate; poly(ε-caprolactone);
53 tissue-engineering bone scaffold; dynamic resorption

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61 **1. Introduction**

62 Additive Manufacturing (AM) has been highly recognised as a promising tool to fabricate
63 patient-specific bone substitutes for the replacement and restoration of lost or irreparable bone
64 tissues due to its unique ability to fully control the complex external shape and internal porous
65 network for the printed scaffold. A virtual scaffold design can be directly generated from the
66 computed tomography (CT) or magnetic resonance imaging (MRI) data for an individual
67 patient. A precise construction of a three-dimensional (3D) structure allows prediction and
68 monitoring of biomechanical behaviour of the scaffold and ensures that it provides sufficient
69 support to the surrounding tissues and dynamic resorption to match the rate of newly-formed
70 tissue ingrowth. Such developments are crucial for the future of the interdisciplinary field of
71 bone tissue engineering.

72 Powder-based inkjet 3D printing is an AM technology that builds a pre-defined structure via
73 depositing binder droplets selectively on powder layers. The selection of powder and binder
74 formulation is the key to a successful 3D printing process as it determines the green strength
75 and structural integrity of a 3D printed (3DP) structure, as well as the reliability of the entire
76 printing system. The 3D printing processability of calcium phosphate (CaP), which is the
77 inorganic constituent of natural bone, has been investigated by introducing different binding
78 approaches. Previous attempts have formulated acidic binders to initiate setting reactions
79 between CaP and acids, or polymeric binders to glue CaP powder via polymer precipitation [1-
80 5]. However, proprietary 3D printing systems are typically equipped with a thermally activated
81 print-head, which only works reliably when water-based inks are used [6-7]. The use of acidic
82 or polymeric binders can significantly compromise the working life and performance of these
83 thermally activated print-heads. To eliminate the use of incompatible binders, some studies
84 have incorporated in-bed binding adhesives into CaP powder bed designed to achieve high
85 reactivity between the powder mixture and the standard water-based binder. Typical binding
86 adhesives that have been used include starch [8], maltodextrin [9], and polyvinyl alcohol [10-
87 11]. However, removal of these binding adhesives during the post-processing stage has resulted
88 in relatively high levels of shrinkage (18 - 20%) [1,12]. A previous study proposed using a
89 highly 3D printable bioceramic material, calcium sulphate (CaSO₄), and a binding adhesive for
90 CaP powder [7]. A setting reaction occurred between CaSO₄ and water-based binder, which
91 solidified surrounding powders and, thus, offered high green strength for the printed structure.

92 Both CaP and CaSO₄ can be resorbed by the host through chemical dissolution and osteoclast
93 resorption processes [13]. However, these two materials have different resorption rates. The
94 resorption of CaSO₄ occurs during the early stages of implantation, often within a month [14-
95 15]. Resorption of CaP occurs at a slower rate than CaSO₄ [16]. The most stable form of CaP,
96 hydroxyapatite (HA), demonstrates a resorption period of 12 – 36 months [17]. Ideally, the
97 resorption rate required for the complete repair of a bone defect should coincide with the natural
98 healing rate [18]. The challenge is that most biomaterials do not solely exhibit a resorption rate
99 that matches the typical rate of bone formation [19]. The concept of combining CaSO₄ and CaP
100 to balance the resorption rate has been proposed in many studies [20-22]. It has been reported
101 that the CaP-CaSO₄ composite exhibited a resorption rate closely matching the rate of new
102 bone formation [19]. Here the rapid dissolution of CaSO₄ during the early stages of
103 implantation created porous spaces favouring cell activity, while HA remained and acted as a
104 support for newly formed tissue [23-24]. The incorporation of HA into CaSO₄ could also
105 enhance osteoconductivity due to its crystalline structure and mineral apatite components
106 analogous to natural bone [19,25].

107 In this study, bioceramic scaffolds were 3DP from a HA-CaSO₄ powder formulation. The
108 printed scaffolds were post-treated using poly(ε-caprolactone) (PCL) to infiltrate the inter-
109 particle spacing; PCL infiltration was demonstrated as a highly efficient method to improve
110 the overall performance of 3DP bioceramic scaffolds [26]. The compressive properties,
111 microstructural morphologies and *in vitro* resorption behaviour, under dynamic flow
112 conditions, of 3DP composite scaffolds were subsequently investigated.

113 **2. Materials and Methods**

114 **2.1. Materials**

115 The combination of 25 wt.% HA (Capital[®], Plasma Biotol Ltd., UK) and 75 wt.% hemihydrate
116 CaSO₄ powder (ZP102, Z Corporation, UK) was used as the powdered material for 3DP
117 manufacturing. The HA powder was milled using a planetary mill (Pulverisette 6, Fristch
118 GmbH, Germany) and then sieved so as to achieve a particle size distribution similar to the
119 hemihydrate CaSO₄ powder, which had D₁₀, D₅₀, and D₉₀ equal to 45.68 μm, 65.35 μm, and
120 94.40 μm, as measured by laser diffraction using a two laser Sympatec HELOS/BF Particle
121 Sizer (Sympatec Ltd, UK). Detailed powder preparation procedures were reported in a previous
122 study [7]. Standard water-based binder (ZB 7, Z Corporation, UK) was used as the binder

123 material. PCL powder (Capa 6506, Perstorp Ltd., UK) was dissolved (12% w/w) in chloroform
124 (288306, Sigma-Aldrich, UK) using an ADS-HP1 hotplate stirrer (Asynt Ltd., UK) at 3000
125 RPM until complete dissolution. Tris-HCl buffer solution (pH = 7.4) was used as the
126 physiological medium during the *in vitro* resorption test.

127 **2.2. Manufacturing process**

128 A cylindrical porous scaffold (diameter and height = 13.2 mm, pore and strut size = 1.2 mm)
129 was designed and manufactured from the HA-CaSO₄ powder using a ZCorp 310 3D printer (Z
130 Corporation, US). The design of an interconnected porous structure can vary greatly according
131 to its overall porosity and the size, shape and arrangement of the pores. Ultimately the ideal
132 design should meet geometrical and mechanical requirement towards a specific bone tissue
133 engineering application in order facilitate effective bone healing. Nevertheless for materials
134 research purposes, a cylindrical shaped specimen of regular porosity and pore architecture and
135 was chosen for this study as it have been previously used in a successful manner [26]. During
136 the 3DP process, each binder droplet was selectively deposited on the print bed to build a cross
137 sectional layer of the structure. The overall structure was 3D-printed consecutively layer by
138 layer — with a layer thickness of 100 μm. After the 3D printing process, the unbound powder
139 was removed using compressed air.

140 Each 3DP scaffold was fully immersed in the PCL solution for 30 s to obtain PCL infiltration.
141 Thereafter, the 3DP scaffold was removed and placed in the fume cupboard under dry, low-
142 pressure conditions for 48 h to facilitate solvent evaporation [26]. The untreated and PCL-
143 treated scaffolds were classified as HA-CaSO₄ and PCL-HA-CaSO₄, respectively.
144 Additionally, CaSO₄ scaffolds were prepared using the same 3DP process and PCL treatment
145 (PCL-CaSO₄) to enable direct comparison during characterisation of the resorption behaviour.

146 **2.3. Compressive properties**

147 Compressive properties of HA-CaSO₄ and PCL-HA-CaSO₄ were determined using a universal
148 materials test system (EZ50, Lloyds Instruments, UK). A 1 kN load cell (XLC 01/2419, Lloyds
149 Instruments, UK) was used and the tests were performed at a rate of displacement of 0.5
150 mm/min. The load cell had a load measurement accuracy of ± 0.5% and could read down to
151 1/200th of its capacity. A total of four scaffolds were tested for each material composition. Each
152 scaffold was tested to failure, which was denoted when the load in the post-peak region had
153 reduced to 80% of the peak load. One thousand force-vs.-displacement data points were then

154 logged for each specimen. The compressive strength was defined as the maximum load
155 recorded, divided by the apparent cross-section area of the scaffold. The compressive modulus
156 was determined by measuring the maximum slope within the linear region of the stress-vs.-
157 strain curve immediately after the toe-in region. Simpson's Rule was used to determine the
158 compressive toughness, which was denoted as the area under the compressive stress-vs.-strain
159 curve to the point of failure.

160 **2.4. Morphology**

161 3DP structures were sectioned longitudinally to facilitate observation of the internal
162 architecture and morphology using field emission scanning electron microscope (SEM) (JEOL
163 JSM-6500F, JEOL Ltd., Japan) at an operating voltage of 5 keV. Each structure was mounted
164 on an aluminium stub using a cold cure resin (Extec Corp, Enfield, CT 06083-1258, US),
165 allowed to cure for 24 ± 2 h and subsequently gold-coated using a sputter chamber prior to
166 SEM examination.

167 **2.5. *In vitro* resorption properties**

168 A high-throughput system was manufactured to facilitate the measurement of the *in vitro*
169 resorption properties of 12 groups of 4 specimens simultaneously (**Figure 1**). Within the
170 system, the physiological medium was circulated using a 12-channel peristaltic pump in an
171 effort to create dynamic flow conditions akin to the human body. Each channel controlled the
172 flow rate of the medium which was connected to each reservoir and chamber. Each reservoir
173 was sealed and placed in a temperature-controlled water bath to maintain the temperature of
174 Tris-HCl buffer solution at 37 ± 1 °C. Each chamber was designed to contain a group of
175 scaffolds ($n = 4$). The buffer solution (240 mL), circulated throughout the system and refreshed
176 weekly, was equally distributed to each 3DP scaffold within the group. It was comparable to
177 the volume of medium (i.e. 60 mL) used in previous static *in vitro* resorption studies [26].

178 The *in vitro* resorption behaviour of HA-CaSO₄, PCL-HA-CaSO₄, and PCL-CaSO₄ under a
179 dynamic flow condition was evaluated using the aforementioned system. The rotational speed
180 of the peristaltic pump was 4.25 RP, which equated to a circulating flow rate of approx. 1
181 mL/min [27]. This flow rate has previously been used for the culturing of engineered bone in
182 perfusion bioreactors [27,28]. Note that the actual flux that 3DP scaffolds experienced may be
183 slightly different from the setup flow rate as it also depended on other factors, such as pore

184 size. The resorption properties were measured over a 56-day period. Every week, a group of
185 scaffolds were removed and characterised.

186 The wet mass was measured after carefully removing all the excess water from the structure
187 using sterile filter paper. Each scaffold was then rinsed with deionised water and dried in an
188 oven at 37 ± 1 °C for 48 h. The dry mass was then measured. The water absorption (wt.%) and
189 mass change (%) before and after immersion in buffer solution were calculated using Equation
190 1 and Equation 2:

$$191 \quad \text{Water absorption (\%)} = \frac{m_{t,w} - m_{t,d}}{m_{t,d}} \times 100 \quad (1)$$

$$192 \quad \text{Mass change (\%)} = \frac{m_{0,d} - m_{t,d}}{m_{0,d}} \times 100 \quad (2)$$

193 where: $m_{0,d}$ = dry mass before immersion in buffer solution (g);

194 $m_{t,d}$ = dry mass after immersion in buffer solution (g);

195 $m_{t,w}$ = wet mass after immersion in buffer solution (g).

196 The compressive properties were also determined for each scaffold following drying as
197 previously described. Although determining the compressive properties of the 3DP structures
198 under wet conditions would have been beneficial as this is more representative of the *in vivo*
199 environment. Notwithstanding this fact, drying each of the 3DP structures prior to mechanical
200 testing was chosen as to allow for the non-destructive characterisation to be undertaken
201 beforehand.

202 X-ray diffraction (XRD) analysis was conducted on the different scaffold types using an X-
203 Pert Pro X-ray diffraction system with an X'Celerator X-ray detector (PANalytical Ltd., UK).
204 X'Pert High Score software was used to identify the phases present in the different scaffold
205 types at Day 0, 7, 28, and 56.

206 Each scaffold type was imaged using SEM to observe the effects of resorption on the
207 microstructure and morphology at Day 0, 7, 28, and 56. X-ray microtomography (μ -CT) was
208 also used to determine the structural evolution of the 3DP structures in the dry condition as a
209 function of *in vitro* resorption using a SkyScan 1174 compact desktop X-ray micro-tomography
210 (μ -CT) scanner system (SkyScan N.V., Belgium). Specifically, scanning was conducted at
211 medium resolution (600 axial μ -CT slices with 1024 x 1024 pixels bitmap image, 16.25 μ m

212 pixel size). The micro-focus X-ray source operated at a voltage of 50 kV and a current of 800
213 μA . Aluminium filters (0.75 mm) were applied for beam hardening reduction. During the
214 scanning process the specimen stage was rotated over 360° at a rotation step of 0.51. At each
215 rotation step an angular shadow projection of the specimen was acquired at an exposure time
216 of 5.5 s. The X-ray shadow projections were then digitised and the acquisition geometry for
217 each scan was extracted from the dataset of transmission images using a reconstruction
218 programme (i.e. smoothing = 4, ring artefact correction = 14 and beam hardening correction =
219 46%). Grayscale values are proportional to the material density [29]. One of the challenges of
220 analysing changes in the structural properties as a function of resorption time was the
221 continuous shift in the X-ray absorption peaks on the grayscale histogram due to density
222 reduction. Therefore, it was difficult to use a same threshold level for materials demonstrating
223 different resorption behaviour. The upper and lower threshold levels for reconstructed cross-
224 sectional images were determined from the grayscale histogram, which was generated using
225 ImageJ software (National Institutes of Health, USA). The threshold levels were selected in
226 positions that best separated materials from empty spaces as well as one material phase from
227 another if distinct peaks were indicated from the histogram. Thereafter, the structural properties
228 of interest (e.g. volume and degree of porosity) were determined using the SkyScan CT-
229 analyser software (Version 1.10.1.0, SkyScan N.V., Belgium). Further morphological analysis
230 on the reconstructed image dataset was performed using *ImageJ* software (National Institutes
231 of Health, USA) with the *BoneJ* plugin [29].

232 **2.6. Statistical analysis**

233 Data collected from each experimental test was evaluated for statistical significance using
234 SPSS 13.0 software (SPSS, USA). Differences between treatment groups were assessed using
235 one-way Analysis of Variance (ANOVA) with post-hoc Bonferroni correction. A p-value less
236 than 0.05 denoted statistical significance. Data analysis was selected on basis of normal
237 probability tests.

238 **3. Results and Discussion**

239 **3.1. Compressive properties**

240 PCL-HA-CaSO₄ exhibited significantly higher compressive properties ($p < 0.05$) than CaP-
241 CaSO₄ (**Figure 2**). One of the major drawbacks associated with scaffold manufacture using
242 powder-based 3D printing technology is poor mechanical performance due to high inter-

243 particle spaces within the printed structures [30]. The resulting stress concentrations yield rapid
244 and catastrophic failure under compressive forces (**Figure 3a**). Compressive loading is the
245 most common mode of loading applied to bone during normal day-to-day activities. Therefore,
246 it is essential to improve the compressive properties of tissue engineered bone scaffolds. The
247 most rational approach to overcome this inherent drawback associated with 3DP scaffolds was
248 to fill the inter-particle spaces using a biocompatible and bioresorbable material following 3D
249 printing. Previous studies have used biopolymers as infiltration materials to improve the
250 mechanical properties of 3DP bioceramic scaffolds [26,31-32]. PCL infiltrating the micropores
251 of the ceramic scaffold increased the interfacial contact area providing a corresponding
252 increase in the compressive properties of the scaffold. In this study, the compressive strength,
253 compressive modulus and toughness of HA-CaSO₄ was significantly increased ($p < 0.001$) from
254 0.84 ± 0.06 MPa, 13.64 ± 1.76 MPa, and $0.03 \pm 0.004 \times 10^{-6}$ J/m³ to 2.64 ± 0.18 MPa (average
255 314% ↑), 63.39 ± 4.92 MPa (average 465% ↑), and $0.26 \pm 0.09 \times 10^{-6}$ J/m³ (average 867% ↑)
256 following PCL infiltration. The infiltrated PCL-HA-CaSO₄ scaffold demonstrated high
257 structural integrity following compressive loading (**Figure 3b**). SEM analysis showed the PCL
258 covered the surface pores and also infiltrated deep into the inter-particle spaces (**Figure 4**).
259 Consequently, the PCL increased the overall density of 3DP scaffolds, connected the adjacent
260 powder particles, and bridged cracks in the form of fibrils during compressive loading.

261 **3.2. *In vitro* resorption properties**

262 The purpose of an *in vitro* resorption experiment was to gain a better understanding of material
263 resorption trends using methods analogous to *in vivo* conditions. Resorption under static
264 conditions is the typical methodology used to evaluate the resorption properties because it is
265 easier to implement. However, implants are not exposed to static conditions *in vivo*, but rather
266 a thermodynamically open environment that involves the dynamic flow of body fluid [33].
267 Therefore, it is more representative to subject 3DP scaffolds to *in vitro* resorption under
268 dynamic flow conditions.

269 In this study, HA-CaSO₄ scaffolds were too fragile to handle following 7 days immersion in
270 the buffer solution. The HA-CaSO₄ scaffold type demonstrated a high specific surface area due
271 to high inter-particle spaces. When subjected to an abrasive attack from the dynamic flow, this
272 resulted in a rapid deterioration of structural properties. In contrast, the PCL-HA-CaSO₄
273 scaffold type maintained increased structural integrity following 56 days of immersion in the

274 buffer solution. Therefore, it was pertinent for the remainder of the study to focus on the
275 behaviour of the PCL-HA-CaSO₄ 3DP scaffold type as oppose to its PCL-CaSO₄ counterpart.

276 PCL-HA-CaSO₄ demonstrated a relatively fast mass loss from Day 0 – 28 ($-48.65 \pm 0.80\%$)
277 (**Figure 5**). Following Day 28, further mass loss continued at a slower rate ($-54.48 \pm 0.54\%$)
278 toward Day 56. PCL-HA-CaSO₄ exhibited significantly lower mass loss than PCL-CaSO₄
279 between Day 28 and Day 56 ($p < 0.05$). The use of continuous dynamic flow removed ceramic
280 dissolution by-products from the 3DP scaffold and reduced the potential for local ion
281 oversaturation. Oversaturation of ions has the potential to confound the resorption process.
282 During the first 28 days of *in vitro* resorption, a strong linear correlation ($R^2 = 0.996$) was
283 observed between mass loss and resorption time, suggesting a constant dissolution rate. The
284 rapid resorption of PCL-HA-CaSO₄ following 28 days of immersion in buffer solution was
285 related to the HA and CaSO₄. It was speculated that ceramics were readily dissolved until a
286 time when the remaining ceramic particles were encapsulated with PCL. Consequently, the rate
287 of further resorption receded.

288 Water absorption of PCL-HA-CaSO₄ also increased rapidly from Day 0 – 28 ($146.36 \pm 4.93\%$)
289 (**Figure 6**). Following Day 28, a gradual reduction in water absorption was observed. After 56
290 days of immersion in buffer solution, the PCL-HA-CaSO₄ scaffold exhibited water absorption
291 of $106.11 \pm 1.92\%$. It was significantly lower ($p < 0.05$) than the PCL-CaSO₄ 3DP scaffold
292 throughout the testing period. Water absorption was a balance between removal of hydrophilic
293 resorbing material and the water uptake of surface flaws, which were created due to material
294 removal. The local regions where ceramic particles were exposed to buffer solution became
295 vulnerable due to dissolution. The dynamic flow demonstrated sufficient shear forces to
296 dislodge ceramic particles from weakened regions, which created cavities on the surface of the
297 3DP scaffold. These cavities are likely to have trapped buffer solution and increased water
298 absorption between Day 0 and Day 28, which represented the period when the fastest mass loss
299 was evident. Consequently, the proportion of PCL increased (**Figure 7**). It is a more
300 hydrophobic material when compared to HA or CaSO₄, and therefore the rate of water
301 absorption of PCL-HA-CaSO₄ slowed after Day 28.

302 The compressive strength of PCL-HA-CaSO₄ decreased by 73.48% from between Day 0 (2.64
303 ± 0.18 MPa) and Day 28 (0.70 ± 0.10 MPa) (**Figure 8**). Thereafter, there was no significant
304 change in the compressive strength until Day 56 (0.50 ± 0.16 MPa). After 7 days of immersion
305 in buffer solution, the compressive modulus rapidly reduced to 37% of the original compressive

306 modulus. However, there was no significant change ($p>0.05$) in the compressive modulus
307 following Day 14. On average, the 3DP scaffold retained 15% of its original compressive
308 modulus after 56 days of immersion in the buffer solution. The compressive properties for
309 PCL-CaSO₄ subjected to resorption followed a similar trend. However, the reduction in both
310 compressive strength and compressive modulus for PCL-CaSO₄ was greater than those for
311 PCL-HA-CaSO₄ after Day 21. The compressive properties were affected by the material
312 removal from the 3DP scaffold structure. The greater resorption between Day 0 and Day 28
313 resulted in a significant reduction in compressive properties ($p<0.05$).

314 PCL was selected as the infiltration material because it can provide long-term mechanical
315 stability to withstand the forces exhibited during the early stages of wound contraction and
316 bone healing owing to its relatively slow resorption profile in contrast to other biopolymers,
317 e.g. poly(lactic-co-glycolic acid) (PLGA) [35]. Additionally, PCL offers better ductile
318 characteristics when compared to poly-L-lactic acid (PPLA), which can help augment the
319 brittle nature of the HA-CaSO₄ and therefore improve its fracture resistance and structural
320 retention during physical loading [26, 36]. An important issue to consider when using a solvent
321 (e.g. chloroform) to dissolve the PCL is to ensure its complete removal before clinical
322 application as this will be critical to the cellular behaviour of the PCL coated 3DP scaffolds.
323 Peroligo et al. (2007) used differential scanning analysis to confirm that full evaporation of the
324 chloroform from a PCL coated bioceramic scaffold structure can be achieved within 24-48 h
325 [37]. They concluded that full evaporation of the chloroform was achieved within this time
326 frame since a relatively thin coating of PCL with a high surface area was used – which is also
327 the case in this investigation. Additionally, Tarafder and Bose (2014) used a solvent to dissolve
328 PCL prior to coating tricalcium phosphate 3DP scaffolds and reported no adverse effect of PCL
329 on the biocompatibility of the scaffold or its *in vivo* bone forming capabilities after six weeks
330 implantation [48].

331 The diffraction peaks of the CaSO₄ from the XRD spectra for the PCL-HA-CaSO₄ attenuated
332 greatly during *in vitro* resorption (**Figure 7**). The CaSO₄ phase (1) was difficult to detect after
333 Day 56 and HA (3) became the dominant inorganic phase detected. The organic phase, PCL
334 (2), also remained within the 3DP scaffold structure, exhibiting diffraction peaks at $2\theta = 21.3^\circ$
335 and 23.7° . This result indicated that CaSO₄ experienced a more intensive level of dissolution
336 when compared to the HA phase. It is well known that CaSO₄ dissolves more rapidly than β -
337 TCP and HA [34]. The loss of scaffold mass was attributed to the steady dissolution of CaSO₄,

338 which was exposed directly to the buffered solution. The remaining CaSO₄ was encapsulated
339 by PCL and therefore, was unable to be detected towards the end of resorption. The absence of
340 CaSO₄ during the latter stages of resorption could provide several benefits: firstly, its rapid
341 dissolution released a large amount of Ca²⁺ ions, which would favour cell proliferation *in vivo*
342 [35-36]. Secondly, the removal of the CaSO₄ left behind empty spaces that could enhance new
343 bone formation [37]. HA remained almost passive because the resorption of multiphase-
344 ceramic always begins with the dissolution of the most soluble phase [37]. The greater
345 dissolution of CaSO₄ created an environment rich in Ca²⁺ ions, which also inhibited HA
346 dissolution. The SEM analysis showed that a large number of ceramic particles remained
347 within the 3DP scaffold at Day 56, which were connected by PCL fibrils (**Figure 9**).
348 Additionally, the scaffold surface was still well covered with the PCL coating layer. Therefore,
349 a high level of structural integrity was still available after the 3DP scaffold was subjected to
350 dynamic *in vitro* resorption for 56 days. This signifies the potential for the 3DP scaffold to
351 offer structural and mechanical support to the newly formed tissues.

352 During the 3D printing process, a solid structure was formed due to the precipitation of ceramic
353 aggregates when the setting reaction between the CaSO₄ powder and water-based binder
354 occurred (**Figure 10**). The aggregated rod-like crystals eroded when the structure was subjected
355 to resorption in the buffer solution. The surface erosion resulted in a smoother topography,
356 which was observed for the PCL-HA-CaSO₄ morphology after 28 days immersion in buffer
357 solution. More significant erosion was demonstrated when the resorption proceeded to Day 56.
358 These results demonstrated that the mechanism of surface erosion was dominant, whereby the
359 material located within the internal structure only resorbed following complete resorption of
360 the external surfaces.

361 One of the challenges of analysing resorption using μ -CT is the continuous shift of the X-ray
362 absorption peaks on the grayscale histogram, which was proportional to the material density
363 [38]. A decrease in material density related to the microporosity created as a consequence of
364 the resorption process. A voxel containing an element of microporosity exhibited a relatively
365 low grayscale value. A more dramatic shift of the X-ray absorption peaks towards lower values
366 was observed on the grayscale histogram of PCL-CaSO₄ 3DP scaffolds compared to their PCL-
367 HA-CaSO₄ counterparts (**Figure 11**). There was no distinct peak observed on the histogram
368 for PCL-CaSO₄ 3DP scaffolds between Day 42 and Day 56. To demonstrate material
369 distribution at different stages of the resorption period, binary images of cross-sections of 3DP

370 scaffolds were generated via thresholding the reconstruction images at different grayscale
371 levels based on the histograms: (1) 24-255 for PCL-HA-CaSO₄ and (2) 5-37 and 37-255 for
372 PCL-CaSO₄ (**Figure 12**). The cross-sectional images of PCL-HA-CaSO₄ after 56 days of
373 resorption did not exhibit a significant change and a high level of structural integrity was
374 maintained. Smaller pores (approx. 100 μm) were created within the structure after 56 days of
375 resorption. However, PCL-CaSO₄ scaffolds experienced a rapid loss of material (**Figure 13**),
376 which correlated to a lower grayscale range at Day 28 as a consequence of the reduction in
377 density throughout the 3DP scaffold. This transformation continued to be more significant as
378 only a few materials were observed on the binary images at Day 56. The dip-coating technique
379 enabled deep infiltration of PCL within the scaffold structure, which efficiently filled the inter-
380 particle spacing. The entire 3DP scaffold structure was maintained due to the PCL
381 demonstrating a low level of resorption by Day 56. Due to this infiltration of PCL within the
382 3DP scaffold it was highly unlikely that large spaces were created unless the *in vitro* resorption
383 reached a relatively high level, which was the case for the PCL-CaSO₄ 3DP scaffold type.
384 However, for the PCL-HA-CaSO₄ based 3DP scaffold, both PCL infiltration and faster
385 dissolution of CaSO₄ reduced HA dissolution. Consequently, a large amount of HA particles
386 remained within the 3DP scaffold structure.

387 This study has been focused on the printability of the polymer coated composite materials and
388 the corresponding *in vitro* data reported over a 56-day period shows promise. Notwithstanding
389 this fact, there may be additional questions that are still unresolved and areas for further
390 development. Although, use of the PCL coating successfully increased the compressive
391 strength to match cancellous bone (2-12 MPa) [47]. The compressive modulus, however, was
392 still lower than that of cancellous bone (100-500 MPa) due to the loose packing nature of
393 ceramic particles. Further improvement in the compressive properties could be achieved
394 through: (1) optimisation of the particle characteristics to achieve more dense powder packing,
395 or (2) incorporation of ceramic particles within the polymer infiltration solution for better
396 structural reinforcement. Further investigation is necessary to understand how the PCL-HA-
397 CaSO₄ based 3DP scaffolds perform under cyclic loading within a pseudo-physiological
398 environment as this is an essential next step before *in vivo* testing using an appropriate animal
399 model. Studies regarding the cytotoxic response of both HA and CaSO₄ based biomaterials
400 have documented favourable findings corroborating their potential in orthopaedic applications
401 [14,25,39-42,48]. However, there may be toxicity issues associated with our composite
402 scaffold since the materials were processed via a novel technology. It is important that

403 compressive and tunable resorption properties of the 3DP scaffolds are appropriately balanced
404 to the sequence of events exhibited during new bone formation *in vivo*, i.e. encouraging new
405 tissue growth during the initial stages of implantation, offering sufficient structural and
406 mechanical support and follow an appropriate resorption profile during the bone healing phase.
407 This balance can be achieved by varying the CaSO₄:CaP:PCL of the final material formulation
408 to mimic the mechanical properties and resorption requirements for a specific clinical
409 application, which will vary considerably depending on number of factors, e.g. the age of
410 patient, bone quality, size and anatomical site of the defect. Further work is currently ongoing
411 to: (1) augment the mechanical properties of the existing 3DP scaffolds and understand how
412 they perform under cyclic loading within a pseudo-physiological environment, (2) determine
413 the extent of biocompatibility and osteogenic properties of these 3DP manufactured scaffolds
414 using an appropriate *in vivo* animal model and (3) understand if the 3DP scaffolds offer the
415 necessary mechanical and structural support and resorption rate required for complete repair of
416 a critical sized bone defect and coincide with sequence of events involved in natural bone
417 healing.

418 **4. Conclusion**

419 This study demonstrated that a blend of HA and CaSO₄ was a reliable formulation for powder-
420 based 3DP technology. The combination of using a fast resorbing CaSO₄ powder and the
421 relatively slower HA powder represents a promising prospect for tuning the bioresorption of
422 3DP scaffolds. The HA-CaSO₄ could be tailored to coincide with tissue growth rate for different
423 surgical procedures. Following manufacture of the 3DP scaffold, PCL infiltration was used to
424 successfully fill the inter-particle spacing within the scaffold and as a result, the compressive
425 properties increased significantly. The *in vitro* resorption properties under a dynamic flow
426 conditions demonstrated PCL inhibited HA dissolution and maintained the overall structural
427 integrity after 56 days of resorption. The remaining scaffold structure accounted for $45.52 \pm$
428 0.54% of the original mass. Rapid reduction in the compressive properties during the first 28
429 days were mainly attributed to CaSO₄ dissolution. The majority of the CaSO₄ powder resorbed
430 within 56 days, while HA remained almost passive. The resorption of the CaSO₄ particles
431 allowed for spaces to become available, which could act as niches, encouraging new tissue
432 growth in the vicinity of the slow resorbing HA. The prepared HA-CaSO₄ scaffold with
433 enhanced mechanical properties and tuneable bioresorption has potential as a good candidate
434 for bone tissue engineering applications.

435 **References**

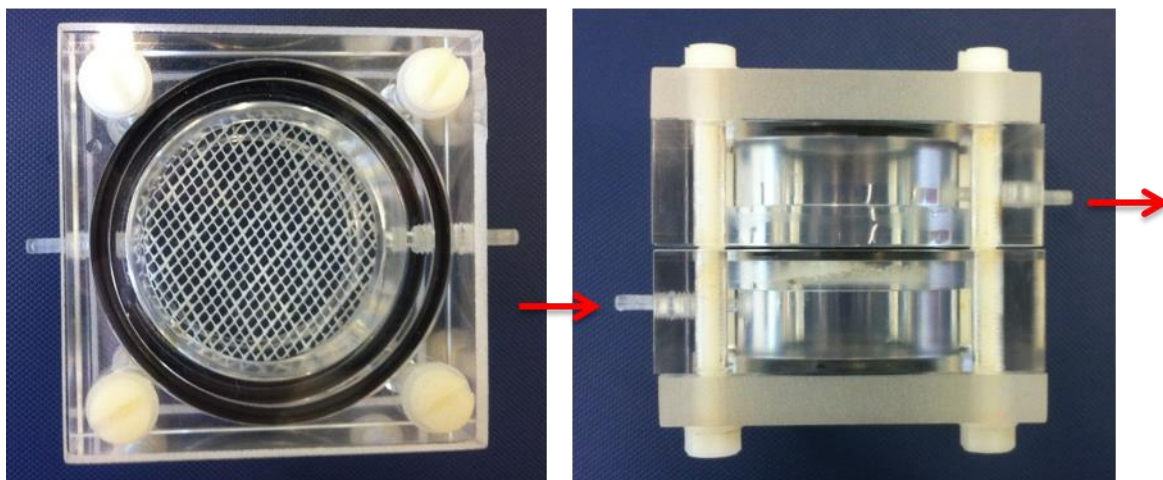
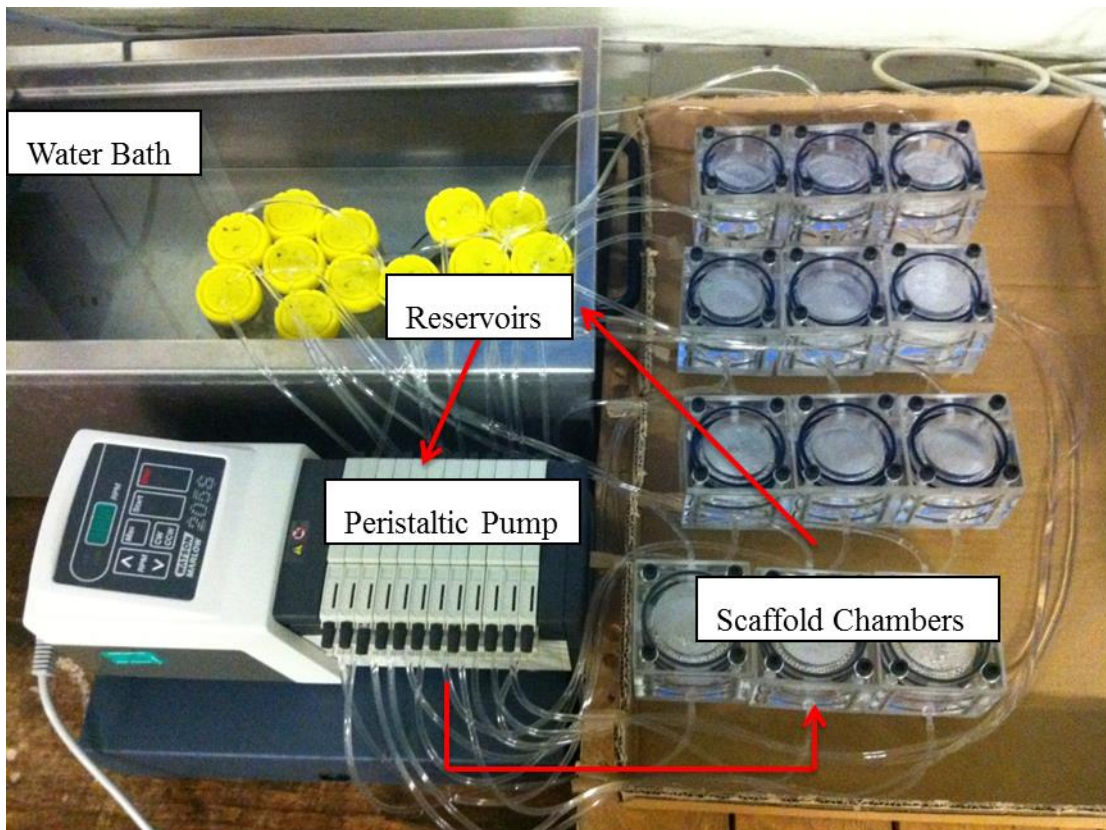
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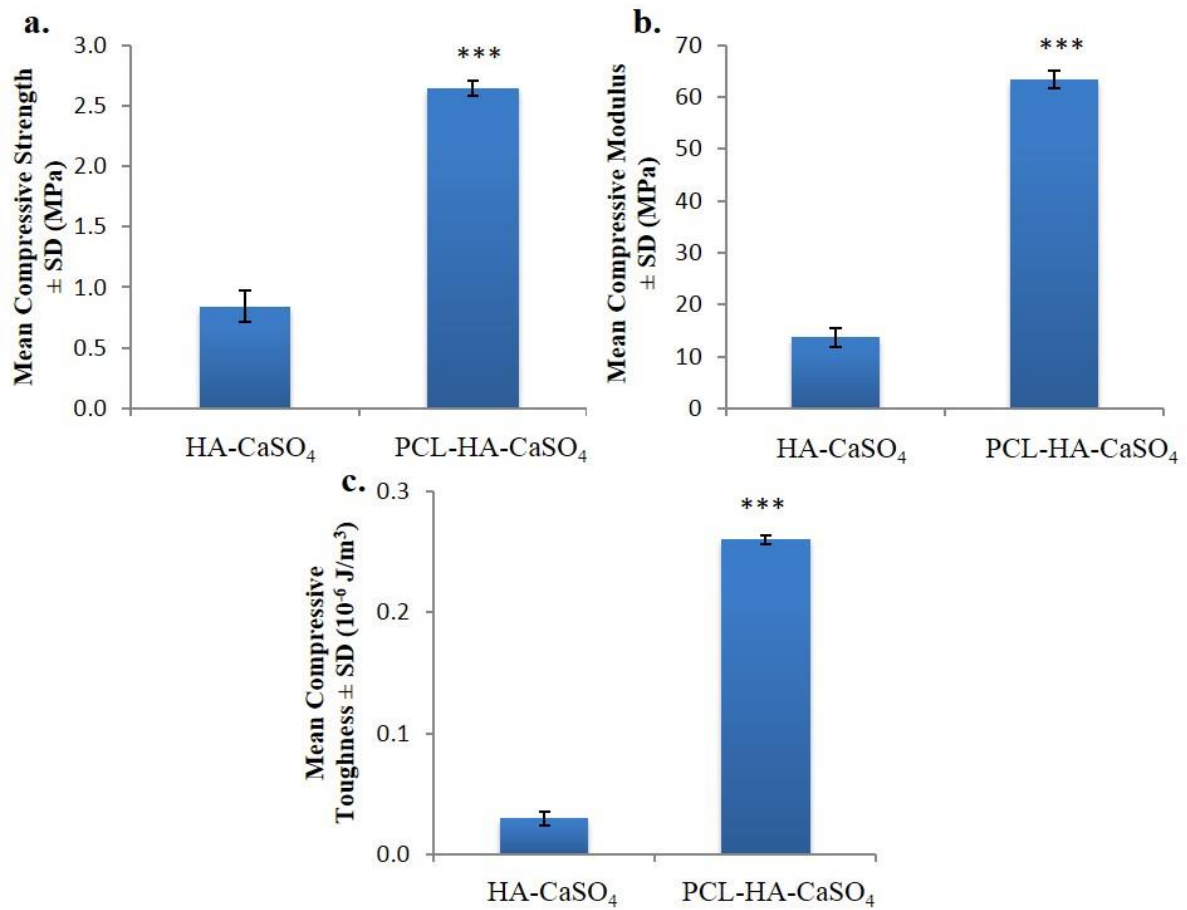
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- 580



581

582 **Figure 1:** (top) Experimental set-up for characterisation of *in vitro* resorption of 3DP scaffolds
 583 under dynamic flow conditions. (bottom) Top and side views of the scaffold chamber.
 584 Direction of fluid flow is indicated by the red arrows.

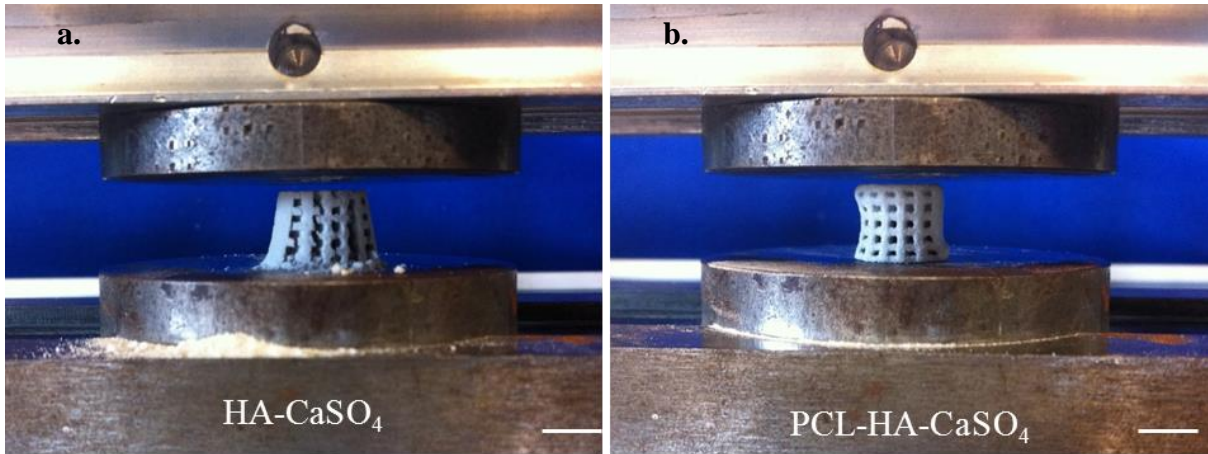
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587 **Figure 2:** (a) Compressive strength, (b) compressive modulus, and (c) compressive toughness
 588 (Mean ± SD) for HA-CaSO₄ and PCL-HA-CaSO₄. ** p < 0.05 and *** p < 0.001, indicating a
 589 significant difference between HA-CaSO₄ and PCL-HA-CaSO₄.

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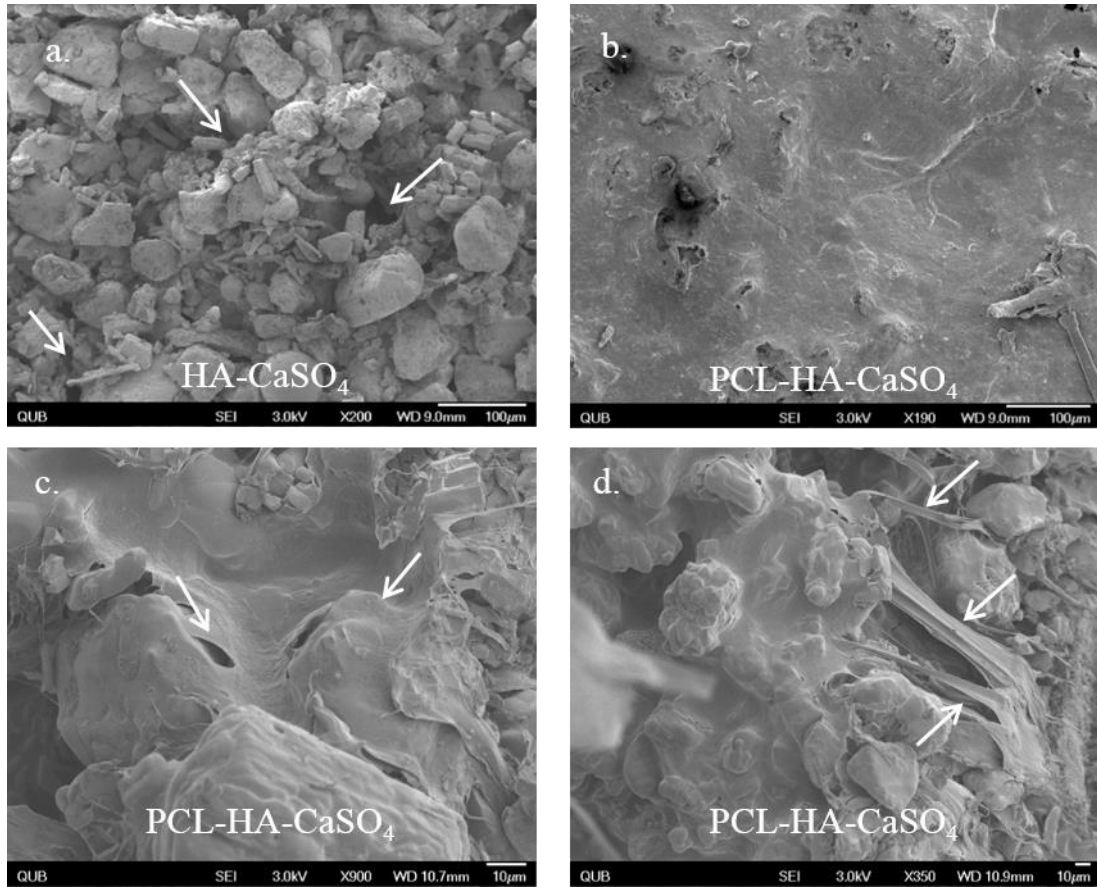


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592 **Figure 3:** Typical compressive deformation for HA-CaSO₄ and PCL-HA-CaSO₄ 3DP scaffolds
593 and difference in scaffold integrity post-failure. Scale bar = 10 mm.

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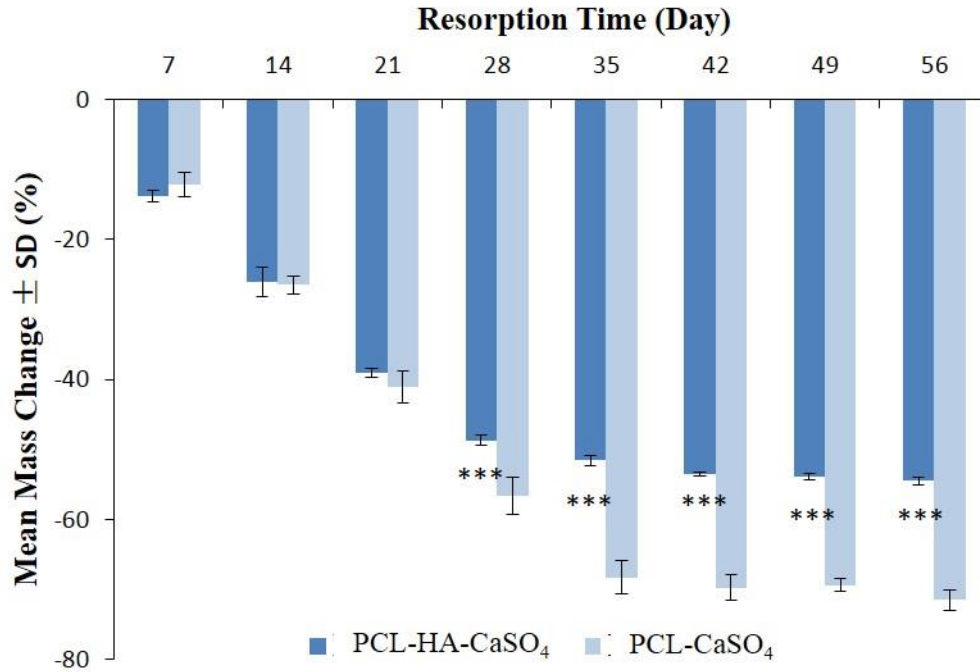


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597 **Figure 4:** SEM images of the surfaces of: (a) HA-CaSO₄, arrows indicate inter-particle gaps
 598 (200x); (b) PCL-HA-CaSO₄ (190x); (c) PCL films filling inter-particle gaps, arrows indicate
 599 PCL coating infiltrating inter-particle gaps (900x); and (d) PCL-HA-CaSO₄ with PCL fibrils
 600 bridging the fractured surface, arrows indicate PCL bridging between ceramic particles within
 601 fractured surface (350x).

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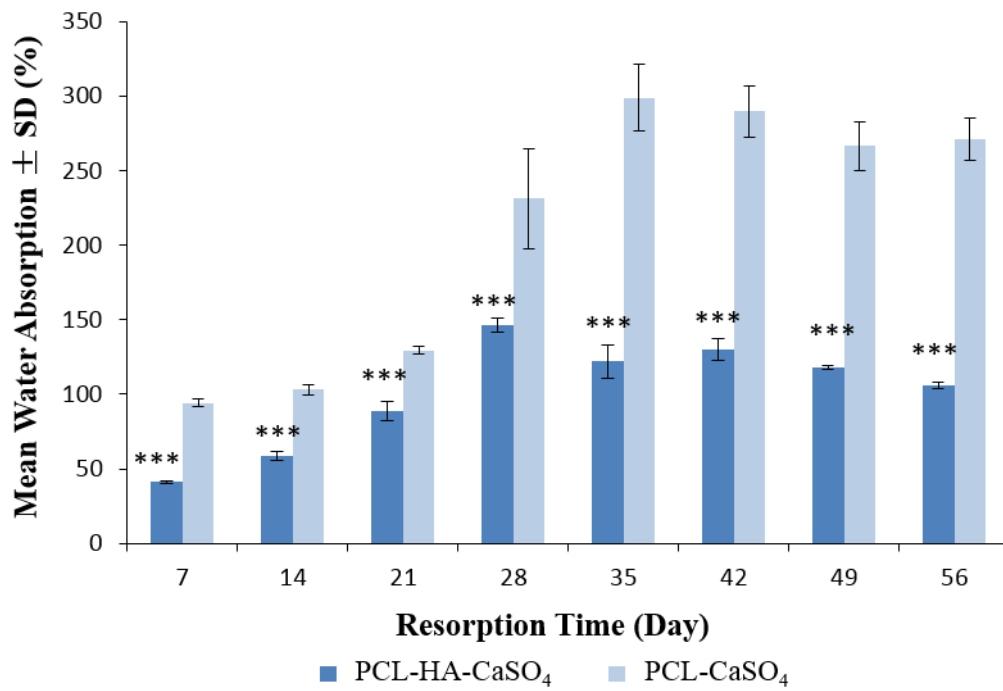
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605 **Figure 5:** Mass change (Mean ± SD) for PCL-HA-CaSO₄ and PCL-CaSO₄ following dynamic
 606 resorption up to 56 days. *** P-value<0.001, indicating a significant difference between PCL-
 607 HA-CaSO₄ and PCL-CaSO₄ at the same time point.

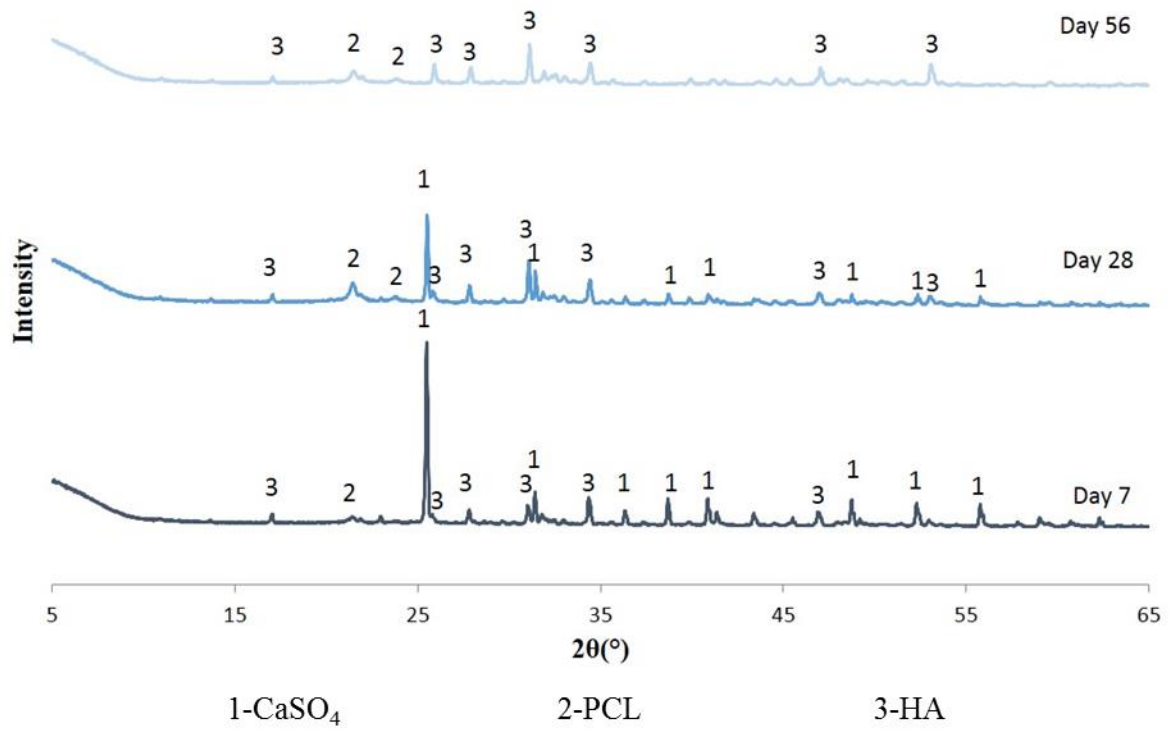
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610 **Figure 6:** Water absorption (Mean ± SD) for PCL-HA-CaSO₄ and PCL-CaSO₄ following
 611 dynamic resorption up to 56 days. ** p-value<0.001, indicating a significant difference
 612 between PCL-HA-CaSO₄ and PCL-CaSO₄ at the same time point.

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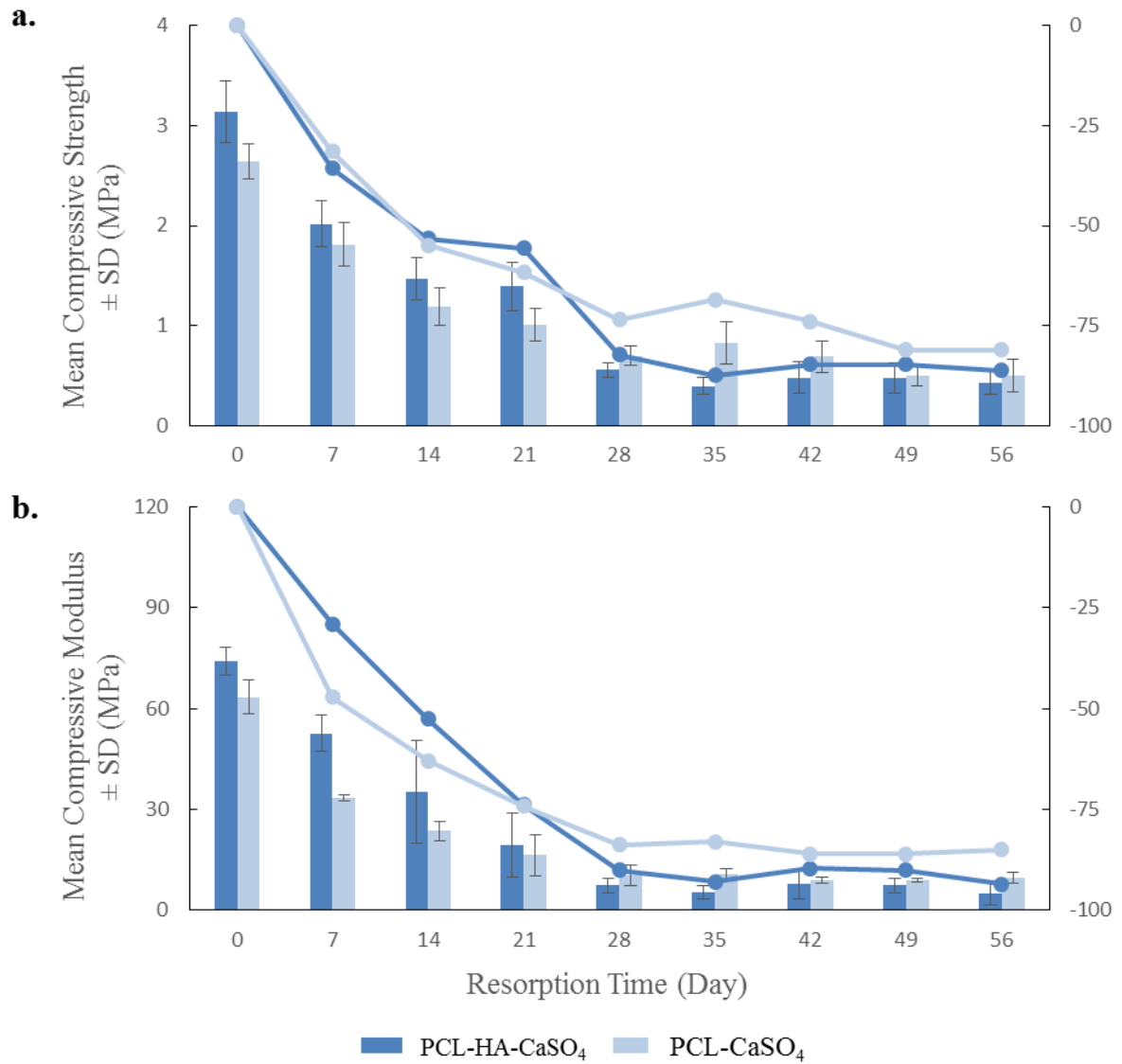


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615 **Figure 7:** XRD spectra for PCL-HA-CaSO₄ after Day 7, 28, and 56 of dynamic *in vitro*
 616 resorption.

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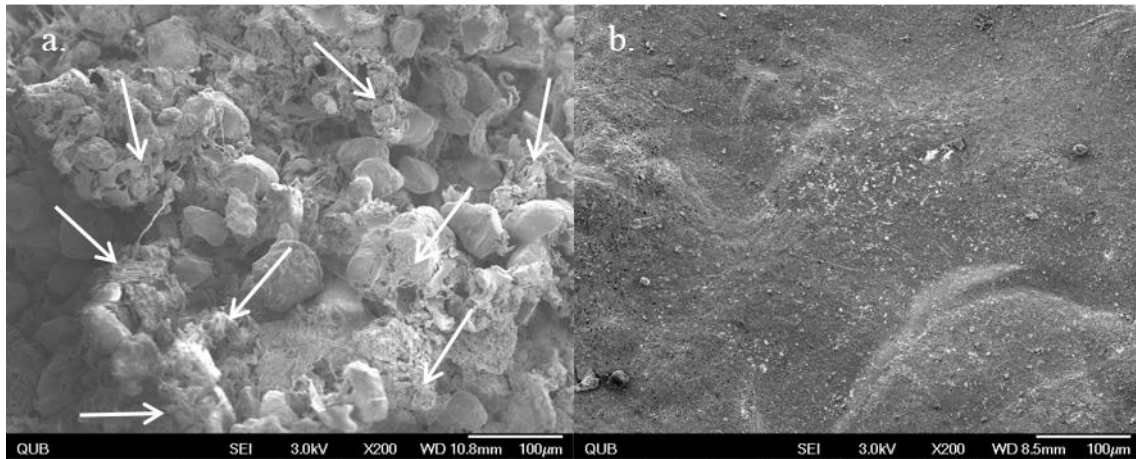
620 **Figure 8:** (a) Compressive strength and (b) compressive modulus (Mean ± SD) for PCL-CaSO₄

621 and PCL-HA-CaSO₄ following resorption up to 56 days. The average percentage change of the

622 properties at each time point is also shown as a line plot referring to the labelled secondary Y-

623 axis.

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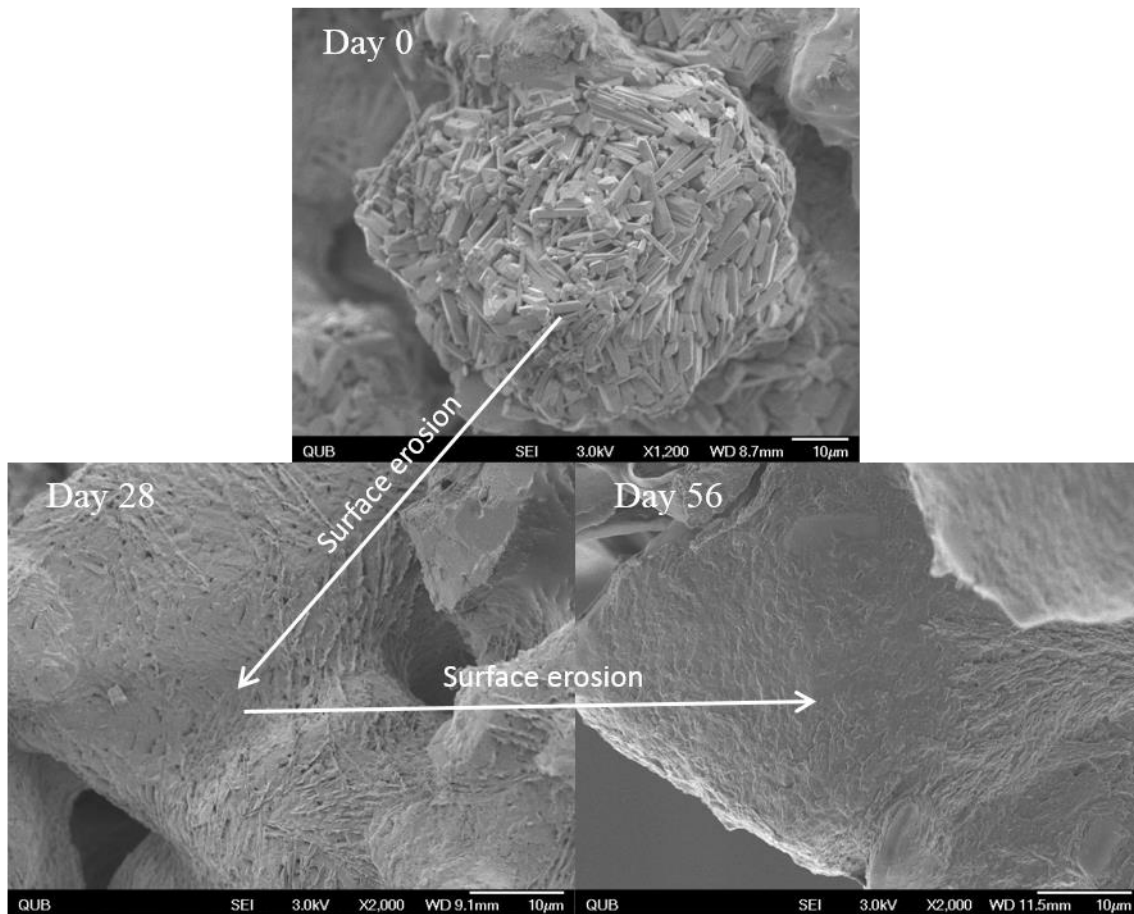


625

626 **Figure 9:** SEM images of (a) the internal structure and (b) the surface of PCL-HA-CaSO₄ after
627 Day 56 of dynamic resorption, arrows indicate PCL fibrils.

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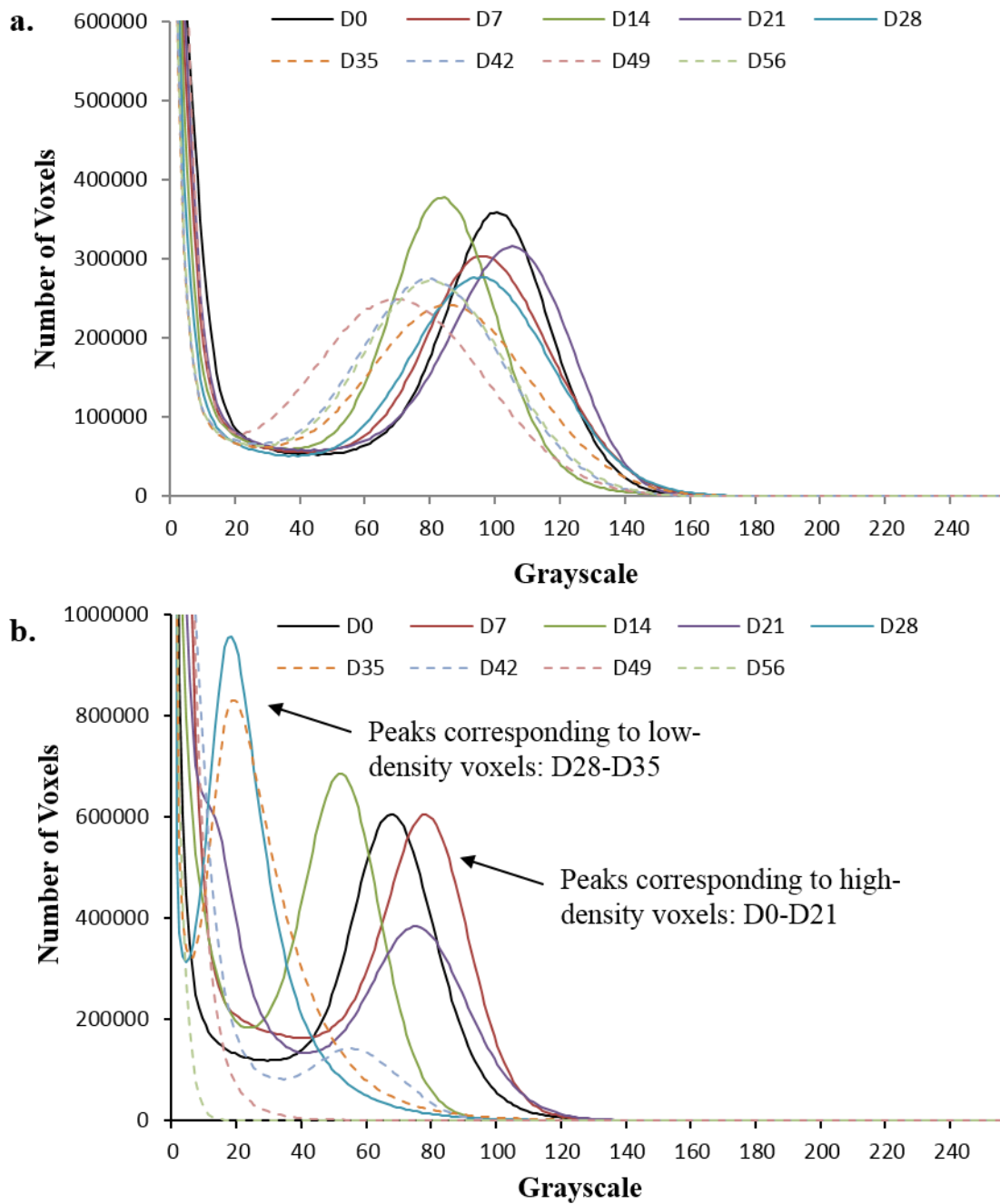


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631 **Figure 10:** SEM images of the ceramic surfaces on the internal structure of PCL-HA-CaSO₄
632 after Day 0, 28, and 56 of dynamic resorption.

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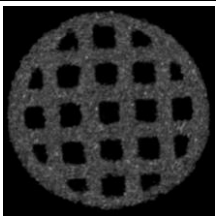
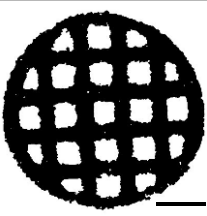
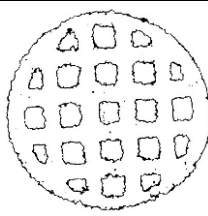
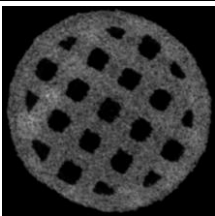
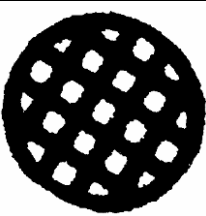
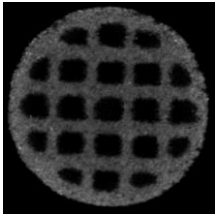

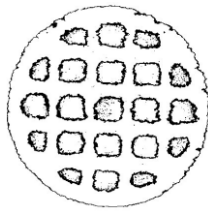
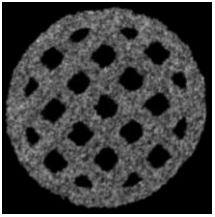
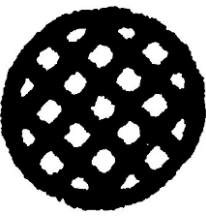
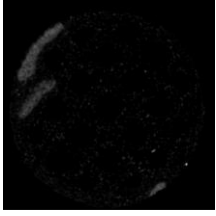

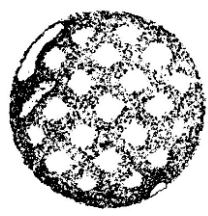
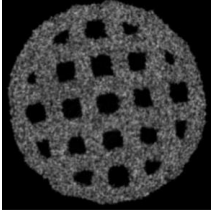
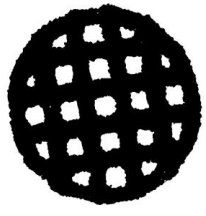
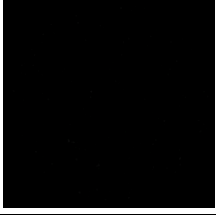

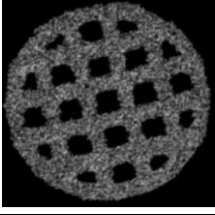

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636 **Figure 11:** Grayscale histograms for μ -CT results of: (a) PCL-HA-CaSO₄ and (b) PCL-CaSO₄
 637 following dynamic resorption up to Day 56.

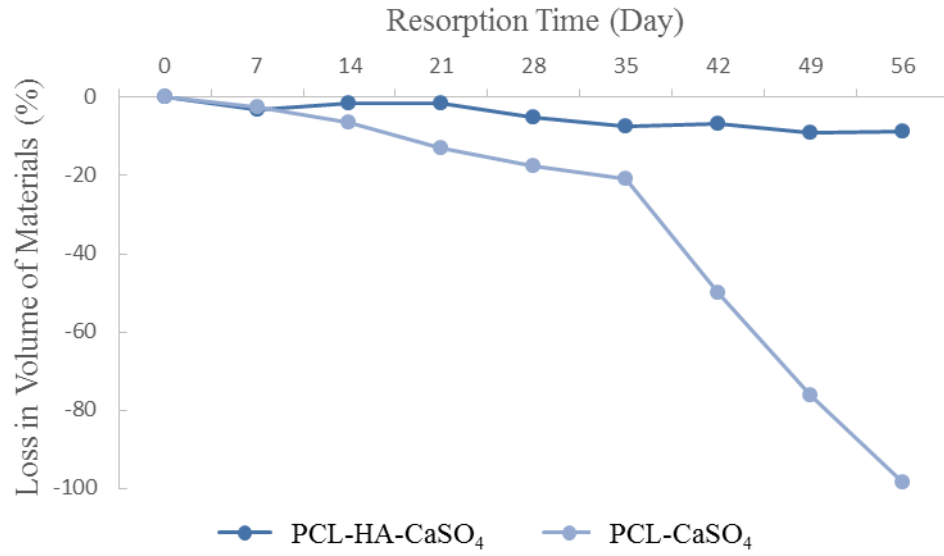
638

| Day | PCL-CaSO ₄ | | | PCL-HA-CaSO ₄ | |
|-----|---|--|---|--|---|
| | Reconstruction Image | Binary Images | | Reconstruction Image | Binary Images |
| | | Grayscale Range: 37-255 | Grayscale Range: 5-37 | | |
| 0 |  |  |  |  |  |
| 7 |  |  |  |  |  |
| 28 |  |  |  |  |  |
| 56 |  | |  |  |  |

639

640 **Figure 12:** Reconstruction and binary images of the cross sections of PCL-HA-CaSO₄ and
641 PCL-CaSO₄ at four time-points of dynamic degradation (Day 0, 7, 28 and 56). The binary
642 images was obtained using thresholding at two levels (37-255 and 5-37) for PCL-CaSO₄ and
643 at one level (24-255) for PCL-HA-CaSO₄. Pixels having grayscale in the range denoted in
644 black. Scale bar = 2 mm.

645



646

647 **Figure 13:** Results determined by μ -CT analysis for material volume loss of PCL-HA-CaSO₄

648 and PCL-CaSO₄ scaffolds as a function of dynamic resorption up to Day 56.

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