

# Development of three-dimensional printing polymer-ceramic scaffolds with enhanced compressive properties and tuneable resorption

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# **Graphical Abstract**



scaffold

# Highlights

- HA:CaSO<sub>4</sub> 3DP scaffolds represent a promising prospect for tuning bioresorption.
- PCL infiltration of the 3DP scaffold significantly improved compressive properties.
- In vitro resorption under dynamic flow demonstrated PCL inhibited HA dissolution.
- CaSO<sub>4</sub> resorption created niches within the scaffold, encouraging new tissue growth.
- PCL provided long-term structural support irrespective of *in vitro* resorption.

# 1 Development of Three-Dimensional Printing Polymer-Ceramic Scaffolds with Enhanced

2 Compressive Properties and Tuneable Resorption

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#### 29 Abstract

30 In this study, bone tissue engineered scaffolds fabricated via powder-based 3D printing from 31 hydroxyapatite (HA) and calcium sulphate (CaSO<sub>4</sub>) powders were investigated. The 32 combination of using a fast resorbing CaSO<sub>4</sub> based powder and the relatively slower HA 33 powder represents a promising prospect for tuning the bioresorption of 3D printed (3DP) 34 scaffolds. These properties could then be tailored to coincide with tissue growth rate for 35 different surgical procedures. The manufactured scaffolds were infiltrated with poly(E-36 caprolactone) (PCL). The PCL infiltrated the inter-particle spacing within the 3DP structures 37 due to the nature of a loosely-packed powder bed and also covered the surface of ceramic-38 based scaffolds. Consequently, the average compressive strength, compressive modulus and 39 toughness increased by 314%, 465% and 867%, respectively. The resorption behaviour of the 40 3DP scaffolds was characterised in vitro using a high-throughput system that mimicked the physiological environment and dynamic flow conditions relevant to the human body. A rapid 41 42 release of CaSO<sub>4</sub> between Day 0 and 28 was commensurate with a reduction in scaffold mass 43 and compressive properties, as well as an increase in medium absorption. In spite of this, HA 44 particles, connected by PCL fibrils, remained within the microstructure after 56 days resorption 45 under dynamic conditions. Consequently, a high level of structural integrity was maintained within the 3DP scaffold. This study presented a porous PCL-HA-CaSO<sub>4</sub> 3DP structure with the 46 47 potential to encourage new tissue growth during the initial stages of implantation and also offering sufficient structural and mechanical support during the bone healing phase. 48

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50

## 51 Keywords

- 52 Additive Manufacturing; 3D printing; hydroxyapatite; calcium sulphate; poly(ε-caprolactone);
- 53 tissue-engineering bone scaffold; dynamic resorption
- 54
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#### 61 **1. Introduction**

Additive Manufacturing (AM) has been highly recognised as a promising tool to fabricate 62 63 patient-specific bone substitutes for the replacement and restoration of lost or irreparable bone 64 tissues due to its unique ability to fully control the complex external shape and internal porous 65 network for the printed scaffold. A virtual scaffold design can be directly generated from the computed tomography (CT) or magnetic resonance imaging (MRI) data for an individual 66 67 patient. A precise construction of a three-dimensional (3D) structure allows prediction and monitoring of biomechanical behaviour of the scaffold and ensures that it provides sufficient 68 69 support to the surrounding tissues and dynamic resorption to match the rate of newly-formed 70 tissue ingrowth. Such developments are crucial for the future of the interdisciplinary field of 71 bone tissue engineering.

72 Powder-based inkjet 3D printing is an AM technology that builds a pre-defined structure via 73 depositing binder droplets selectively on powder layers. The selection of powder and binder 74 formulation is the key to a successful 3D printing process as it determines the green strength 75 and structural integrity of a 3D printed (3DP) structure, as well as the reliability of the entire 76 printing system. The 3D printing processability of calcium phosphate (CaP), which is the 77 inorganic constituent of natural bone, has been investigated by introducing different binding 78 approaches. Previous attempts have formulated acidic binders to initiate setting reactions 79 between CaP and acids, or polymeric binders to glue CaP powder via polymer precipitation [1-80 5]. However, proprietary 3D printing systems are typically equipped with a thermally activated 81 print-head, which only works reliably when water-based inks are used [6-7]. The use of acidic 82 or polymeric binders can significantly compromise the working life and performance of these 83 thermally activated print-heads. To eliminate the use of incompatible binders, some studies 84 have incorporated in-bed binding adhesives into CaP powder bed designed to achieve high 85 reactivity between the powder mixture and the standard water-based binder. Typical binding 86 adhesives that have been used include starch [8], maltodextrin [9], and polyvinyl alcohol [10-87 11]. However, removal of these binding adhesives during the post-processing stage has resulted 88 in relatively high levels of shrinkage (18 - 20%) [1,12]. A previous study proposed using a 89 highly 3D printable bioceramic material, calcium sulphate (CaSO<sub>4</sub>), and a binding adhesive for 90 CaP powder [7]. A setting reaction occurred between CaSO<sub>4</sub> and water-based binder, which 91 solidified surrounding powders and, thus, offered high green strength for the printed structure.

92 Both CaP and CaSO<sub>4</sub> can be resorbed by the host through chemical dissolution and osteoclast 93 resorption processes [13]. However, these two materials have different resorption rates. The 94 resorption of CaSO<sub>4</sub> occurs during the early stages of implantation, often within a month [14-95 15]. Resorption of CaP occurs at a slower rate than CaSO<sub>4</sub> [16]. The most stable form of CaP, 96 hydroxyapatite (HA), demonstrates a resorption period of 12 - 36 months [17]. Ideally, the 97 resorption rate required for the complete repair of a bone defect should coincide with the natural 98 healing rate [18]. The challenge is that most biomaterials do not solely exhibit a resorption rate 99 that matches the typical rate of bone formation [19]. The concept of combining CaSO<sub>4</sub> and CaP 100 to balance the resorption rate has been proposed in many studies [20-22]. It has been reported 101 that the CaP-CaSO<sub>4</sub> composite exhibited a resorption rate closely matching the rate of new 102 bone formation [19]. Here the rapid dissolution of CaSO<sub>4</sub> during the early stages of 103 implantation created porous spaces favouring cell activity, while HA remained and acted as a 104 support for newly formed tissue [23-24]. The incorporation of HA into CaSO<sub>4</sub> could also 105 enhance osteoconductivity due to its crystalline structure and mineral apatite components 106 analogous to natural bone [19,25].

In this study, bioceramic scaffolds were 3DP from a HA-CaSO<sub>4</sub> powder formulation. The printed scaffolds were post-treated using  $poly(\varepsilon$ -caprolactone) (PCL) to infiltrate the interparticle spacing; PCL infiltration was demonstrated as a highly efficient method to improve the overall performance of 3DP bioceramic scaffolds [26]. The compressive properties, microstructural morphologies and *in vitro* resorption behaviour, under dynamic flow conditions, of 3DP composite scaffolds were subsequently investigated.

#### 113 **2. Materials and Methods**

### 114 **2.1. Materials**

The combination of 25 wt.% HA (Capital<sup>®</sup>, Plasma Biotal Ltd., UK) and 75 wt.% hemihydrate 115 CaSO<sub>4</sub> powder (ZP102, Z Corporation, UK) was used as the powdered material for 3DP 116 117 manufacturing. The HA powder was milled using a planetary mill (Pulverisette 6, Fristch GmbH, Germany) and then sieved so as to achieve a particle size distribution similar to the 118 119 hemihydrate CaSO<sub>4</sub> powder, which had D<sub>10</sub>, D<sub>50</sub>, and D<sub>90</sub> equal to 45.68 µm, 65.35 µm, and 120 94.40 µm, as measured by laser diffraction using a two laser Sympatec HELOS/BF Particle 121 Sizer (Sympatec Ltd, UK). Detailed powder preparation procedures were reported in a previous 122 study [7]. Standard water-based binder (ZB 7, Z Corporation, UK) was used as the binder

- 123 material. PCL powder (Capa 6506, Perstorp Ltd., UK) was dissolved (12% w/w) in chloroform
- 124 (288306, Sigma-Aldrich, UK) using an ADS-HP1 hotplate stirrer (Asynt Ltd., UK) at 3000
- 125 RPM until complete dissolution. Tris-HCl buffer solution (pH = 7.4) was used as the
- 126 physiological medium during the *in vitro* resorption test.

#### 127 **2.2. Manufacturing process**

128 A cylindrical porous scaffold (diameter and height = 13.2 mm, pore and strut size = 1.2 mm) 129 was designed and manufactured from the HA-CaSO<sub>4</sub> powder using a ZCorp 310 3D printer (Z 130 Corporation, US). The design of an interconnected porous structure can vary greatly according 131 to its overall porosity and the size, shape and arrangement of the pores. Ultimately the ideal 132 design should meet geometrical and mechanical requirement towards a specific bone tissue 133 engineering application in order facilitate effective bone healing. Nevertheless for materials 134 research purposes, a cylindrical shaped specimen of regular porosity and pore architecture and 135 was chosen for this study as it have been previously used in a successful manner [26]. During 136 the 3DP process, each binder droplet was selectively deposited on the print bed to build a cross 137 sectional layer of the structure. The overall structure was 3D-printed consecutively layer by layer — with a layer thickness of 100 µm. After the 3D printing process, the unbound powder 138 139 was removed using compressed air.

Each 3DP scaffold was fully immersed in the PCL solution for 30 s to obtain PCL infiltration. Thereafter, the 3DP scaffold was removed and placed in the fume cupboard under dry, lowpressure conditions for 48 h to facilitate solvent evaporation [26]. The untreated and PCLtreated scaffolds were classified as HA-CaSO<sub>4</sub> and PCL-HA-CaSO<sub>4</sub>, respectively. Additionally, CaSO<sub>4</sub> scaffolds were prepared using the same 3DP process and PCL treatment (PCL-CaSO<sub>4</sub>) to enable direct comparison during characterisation of the resorption behaviour.

#### 146 **2.3. Compressive properties**

147 Compressive properties of HA-CaSO<sub>4</sub> and PCL-HA-CaSO<sub>4</sub> were determined using a universal 148 materials test system (EZ50, Lloyds Instruments, UK). A 1 kN load cell (XLC 01/2419, Lloyds 149 Instruments, UK) was used and the tests were performed at a rate of displacement of 0.5 150 mm/min. The load cell had a load measurement accuracy of  $\pm$  0.5% and could read down to 151 1/200<sup>th</sup> of its capacity. A total of four scaffolds were tested for each material composition. Each 152 scaffold was tested to failure, which was denoted when the load in the post-peak region had 153 reduced to 80% of the peak load. One thousand force-vs.-displacement data points were then 154 logged for each specimen. The compressive strength was defined as the maximum load 155 recorded, divided by the apparent cross-section area of the scaffold. The compressive modulus 156 was determined by measuring the maximum slope within the linear region of the stress-vs.-157 strain curve immediately after the toe-in region. Simpson's Rule was used to determine the 158 compressive toughness, which was denoted as the area under the compressive stress-vs.-strain 159 curve to the point of failure.

## 160 **2.4. Morphology**

161 3DP structures were sectioned longitudinally to facilitate observation of the internal 162 architecture and morphology using field emission scanning electron microscope (SEM) (JEOL 163 JSM-6500F, JEOL Ltd., Japan) at an operating voltage of 5 keV. Each structure was mounted 164 on an aluminium stub using a cold cure resin (Extec Corp, Enfield, CT 06083-1258, US), 165 allowed to cure for  $24 \pm 2$  h and subsequently gold-coated using a sputter chamber prior to 166 SEM examination.

#### 167 **2.5.** *In vitro* resorption properties

168 A high-throughput system was manufactured to facilitate the measurement of the *in vitro* 169 resorption properties of 12 groups of 4 specimens simultaneously (Figure 1). Within the 170 system, the physiological medium was circulated using a 12-channel peristaltic pump in an 171 effort to create dynamic flow conditions akin to the human body. Each channel controlled the 172 flow rate of the medium which was connected to each reservoir and chamber. Each reservoir 173 was sealed and placed in a temperature-controlled water bath to maintain the temperature of Tris-HCl buffer solution at  $37 \pm 1$  °C. Each chamber was designed to contain a group of 174 175 scaffolds (n = 4). The buffer solution (240 mL), circulated throughout the system and refreshed 176 weekly, was equally distributed to each 3DP scaffold within the group. It was comparable to 177 the volume of medium (i.e. 60 mL) used in previous static *in vitro* resorption studies [26].

The *in vitro* resorption behaviour of HA-CaSO<sub>4</sub>, PCL-HA-CaSO<sub>4</sub>, and PCL-CaSO<sub>4</sub> under a dynamic flow condition was evaluated using the aforementioned system. The rotational speed of the peristaltic pump was 4.25 RP, which equated to a circulating flow rate of approx. 1 mL/min [27]. This flow rate has previously been used for the culturing of engineered bone in perfusion bioreactors [27,28]. Note that the actual flux that 3DP scaffolds experienced may be slightly different from the setup flow rate as it also depended on other factors, such as pore size. The resorption properties were measured over a 56-day period. Every week, a group ofscaffolds were removed and characterised.

The wet mass was measured after carefully removing all the excess water from the structure using sterile filter paper. Each scaffold was then rinsed with deionised water and dried in an oven at  $37 \pm 1$  °C for 48 h. The dry mass was then measured. The water absorption (wt.%) and mass change (%) before and after immersion in buffer solution were calculated using Equation 1 and Equation 2:

191 
$$Water absorption (\%) = \frac{m_{t,w} - m_{t,d}}{m_{t,d}} \times 100$$
(1)

192 
$$Mass change (\%) = \frac{m_{0,d} - m_{t,d}}{m_{0,d}} \times 100$$
 (2)

193 where:  $m_{0,d} = dry$  mass before immersion in buffer solution (g);

194 
$$m_{t,d} = dry mass after immersion in buffer solution (g);$$

195 
$$m_{t,w}$$
 = wet mass after immersion in buffer solution (g).

The compressive properties were also determined for each scaffold following drying as previously described. Although determining the compressive properties of the 3DP structures under wet conditions would have been beneficial as this is more representative of the *in vivo* environment. Notwithstanding this fact, drying each of the 3DP structures prior to mechanical testing was chosen as to allow for the non-destructive characterisation to be undertaken beforehand.

X-ray diffraction (XRD) analysis was conducted on the different scaffold types using an XPert Pro X-ray diffraction system with an X'Celerator X-ray detector (PANalytical Ltd., UK).
X'Pert High Score software was used to identify the phases present in the different scaffold

205 types at Day 0, 7, 28, and 56.

Each scaffold type was imaged using SEM to observe the effects of resorption on the microstructure and morphology at Day 0, 7, 28, and 56. X-ray microtomography ( $\mu$ -CT) was also used to determine the structural evolution of the 3DP structures in the dry condition as a function of *in vitro* resorption using a SkyScan 1174 compact desktop X-ray micro-tomography ( $\mu$ -CT) scanner system (SkyScan N.V., Belgium). Specifically, scanning was conducted at medium resolution (600 axial  $\mu$ -CT slices with 1024 x 1024 pixels bitmap image, 16.25  $\mu$ m 212 pixel size). The micro-focus X-ray source operated at a voltage of 50 kV and a current of 800 213  $\mu$ A. Aluminium filters (0.75 mm) were applied for beam hardening reduction. During the 214 scanning process the specimen stage was rotated over 360° at a rotation step of 0.51. At each 215 rotation step an angular shadow projection of the specimen was acquired at an exposure time 216 of 5.5 s. The X- ray shadow projections were then digitised and the acquisition geometry for 217 each scan was extracted from the dataset of transmission images using a reconstruction 218 programme (i.e. smoothing = 4, ring artefact correction = 14 and beam hardening correction = 219 46%). Grayscales values are proportional to the material density [29]. One of the challenges of 220 analysing changes in the structural properties as a function of resorption time was the 221 continuous shift in the X-ray absorption peaks on the grayscale histogram due to density 222 reduction. Therefore, it was difficult to use a same threshold level for materials demonstrating 223 different resorption behaviour. The upper and lower threshold levels for reconstructed cross-224 sectional images were determined from the grayscale histogram, which was generated using 225 ImageJ software (National Institutes of Health, USA). The threshold levels were selected in 226 positions that best separated materials from empty spaces as well as one material phase from 227 another if distinct peaks were indicated from the histogram. Thereafter, the structural properties 228 of interest (e.g. volume and degree of porosity) were determined using the SkyScan CT-229 analyser software (Version1.10.1.0, SkyScanN.V., Belgium). Further morphological analysis 230 on the reconstructed image dataset was performed using *ImageJ* software (National Institutes 231 of Health, USA) with the BoneJ plugin [29].

#### 232 **2.6. Statistical analysis**

Data collected from each experimental test was evaluated for statistical significance using SPSS13.0 software (SPSS, USA). Differences between treatment groups were assessed using one-way Analysis of Variance (ANOVA) with post-hoc Bonferroni correction. A p-value less than 0.05 denoted statistical significance. Data analysis was selected on basis of normal probability tests.

238 **3. Results and Discussion** 

## 239 **3.1.** Compressive properties

240 PCL-HA-CaSO<sub>4</sub> exhibited significantly higher compressive properties (p<0.05) than CaP-241 CaSO<sub>4</sub> (**Figure 2**). One of the major drawbacks associated with scaffold manufacture using 242 powder-based 3D printing technology is poor mechanical performance due to high inter243 particle spaces within the printed structures [30]. The resulting stress concentrations yield rapid 244 and catastrophic failure under compressive forces (Figure 3a). Compressive loading is the 245 most common mode of loading applied to bone during normal day-to-day activities. Therefore, 246 it is essential to improve the compressive properties of tissue engineered bone scaffolds. The 247 most rational approach to overcome this inherent drawback associated with 3DP scaffolds was 248 to fill the inter-particle spaces using a biocompatible and bioresorbable material following 3D 249 printing. Previous studies have used biopolymers as infiltration materials to improve the 250 mechanical properties of 3DP bioceramic scaffolds [26,31-32]. PCL infiltrating the micropores 251 of the ceramic scaffold increased the interfacial contact area providing a corresponding 252 increase in the compressive properties of the scaffold. In this study, the compressive strength, 253 compressive modulus and toughness of HA-CaSO<sub>4</sub> was significantly increased (p<0.001) from  $0.84 \pm 0.06$  MPa,  $13.64 \pm 1.76$  MPa, and  $0.03 \pm 0.004$   $10^{-6}$  J/m<sup>3</sup> to  $2.64 \pm 0.18$  MPa (average 254 255 314%  $\uparrow$ ), 63.39 ± 4.92 MPa (average 465%  $\uparrow$ ), and 0.26 ± 0.09 10<sup>-6</sup> J/m<sup>3</sup> (average 867%  $\uparrow$ ) 256 following PCL infiltration. The infiltrated PCL-HA-CaSO<sub>4</sub> scaffold demonstrated high 257 structural integrity following compressive loading (Figure 3b). SEM analysis showed the PCL 258 covered the surface pores and also infiltrated deep into the inter-particle spaces (Figure 4). 259 Consequently, the PCL increased the overall density of 3DP scaffolds, connected the adjacent 260 powder particles, and bridged cracks in the form of fibrils during compressive loading.

## 261 **3.2.** *In vitro* resorption properties

The purpose of an *in vitro* resorption experiment was to gain a better understanding of material resorption trends using methods analogous to *in vivo* conditions. Resorption under static conditions is the typical methodology used to evaluate the resorption properties because it is easier to implement. However, implants are not exposed to static conditions *in vivo*, but rather a thermodynamically open environment that involves the dynamic flow of body fluid [33]. Therefore, it is more representative to subject 3DP scaffolds to *in vitro* resorption under dynamic flow conditions.

In this study, HA-CaSO<sub>4</sub> scaffolds were too fragile to handle following 7 days immersion in the buffer solution. The HA-CaSO<sub>4</sub> scaffold type demonstrated a high specific surface area due to high inter-particle spaces. When subjected to an abrasive attack from the dynamic flow, this resulted in a rapid deterioration of structural properties. In contrast, the PCL-HA-CaSO<sub>4</sub> scaffold type maintained increased structural integrity following 56 days of immersion in the buffer solution. Therefore, it was pertinent for the remainder of the study to focus on the
behaviour of the PCL-HA-CaSO<sub>4</sub> 3DP scaffold type as oppose to its PCL-CaSO<sub>4</sub> counterpart.

276 PCL-HA-CaSO<sub>4</sub> demonstrated a relatively fast mass loss from Day 0 - 28 (-48.65  $\pm$  0.80%) 277 (Figure 5). Following Day 28, further mass loss continued at a slower rate  $(-54.48 \pm 0.54\%)$ 278 toward Day 56. PCL-HA-CaSO<sub>4</sub> exhibited significantly lower mass loss than PCL-CaSO<sub>4</sub> 279 between Day 28 and Day 56 (p<0.05). The use of continuous dynamic flow removed ceramic 280 dissolution by-products from the 3DP scaffold and reduced the potential for local ion 281 oversaturation. Oversaturation of ions has the potential to confound the resorption process. 282 During the first 28 days of *in vitro* resorption, a strong linear correlation ( $R^2 = 0.996$ ) was 283 observed between mass loss and resorption time, suggesting a constant dissolution rate. The 284 rapid resorption of PCL-HA-CaSO<sub>4</sub> following 28 days of immersion in buffer solution was 285 related to the HA and CaSO<sub>4</sub>. It was speculated that ceramics were readily dissolved until a 286 time when the remaining ceramic particles were encapsulated with PCL. Consequently, the rate 287 of further resorption receded.

288 Water absorption of PCL-HA-CaSO<sub>4</sub> also increased rapidly from Day 0 - 28 (146.36 ± 4.93%) 289 (Figure 6). Following Day 28, a gradual reduction in water absorption was observed. After 56 290 days of immersion in buffer solution, the PCL-HA-CaSO<sub>4</sub> scaffold exhibited water absorption 291 of 106.11  $\pm$  1.92%. It was significantly lower (p<0.05) than the PCL-CaSO<sub>4</sub> 3DP scaffold throughout the testing period. Water absorption was a balance between removal of hydrophilic 292 293 resorbing material and the water uptake of surface flaws, which were created due to material removal. The local regions where ceramic particles were exposed to buffer solution became 294 295 vulnerable due to dissolution. The dynamic flow demonstrated sufficient shear forces to 296 dislodge ceramic particles from weakened regions, which created cavities on the surface of the 297 3DP scaffold. These cavities are likely to have trapped buffer solution and increased water 298 absorption between Day 0 and Day 28, which represented the period when the fastest mass loss 299 was evident. Consequently, the proportion of PCL increased (Figure 7). It is a more 300 hydrophobic material when compared to HA or CaSO<sub>4</sub>, and therefore the rate of water 301 absorption of PCL-HA-CaSO<sub>4</sub> slowed after Day 28.

302 The compressive strength of PCL-HA-CaSO<sub>4</sub> decreased by 73.48% from between Day 0 (2.64

 $\pm 0.18$ MPa) and Day 28 (0.70  $\pm 0.10$ MPa) (Figure 8). Thereafter, there was no significant

- 304 change in the compressive strength until Day 56 ( $0.50 \pm 0.16$  MPa). After 7 days of immersion
- in buffer solution, the compressive modulus rapidly reduced to 37% of the original compressive

306 modulus. However, there was no significant change (p>0.05) in the compressive modulus 307 following Day 14. On average, the 3DP scaffold retained 15% of its original compressive 308 modulus after 56 days of immersion in the buffer solution. The compressive properties for 309 PCL-CaSO<sub>4</sub> subjected to resorption followed a similar trend. However, the reduction in both 310 compressive strength and compressive modulus for PCL-CaSO<sub>4</sub> was greater than those for 311 PCL-HA-CaSO<sub>4</sub> after Day 21. The compressive properties were affected by the material 312 removal from the 3DP scaffold structure. The greater resorption between Day 0 and Day 28 313 resulted in a significant reduction in compressive properties (p < 0.05).

314 PCL was selected as the infiltration material because it can provide long-term mechanical 315 stability to withstand the forces exhibited during the early stages of wound contraction and 316 bone healing owing to its relatively slow resorption profile in contrast to other biopolymers, 317 e.g. poly(lactic-co-glycolic acid) (PLGA) [35]. Additionally, PCL offers better ductile characteristics when compared to poly-L-lactic acid (PPLA), which can help augment the 318 319 brittle nature of the HA-CaSO<sub>4</sub> and therefore improve its fracture resistance and structural 320 retention during physical loading [26, 36]. An important issue to consider when using a solvent 321 (e.g. chloroform) to dissolve the PCL is to ensure its complete removal before clinical 322 application as this will be critical to the cellular behaviour of the PCL coated 3DP scaffolds. 323 Peroligo et al. (2007) used differential scanning analysis to confirm that full evaporation of the 324 chloroform from a PCL coated bioceramic scaffold structure can be achieved within 24-48 h 325 [37]. They concluded that full evaporation of the chloroform was achieved within this time frame since a relatively thin coating of PCL with a high surface area was used – which is also 326 327 the case in this investigation. Additionally, Tarafder and Bose (2014) used a solvent to dissolve 328 PCL prior to coating tricalcium phosphate 3DP scaffolds and reported no adverse effect of PCL 329 on the biocompatibility of the scaffold or its *in vivo* bone forming capabilities after six weeks 330 implantation [48].

The diffraction peaks of the CaSO<sub>4</sub> from the XRD spectra for the PCL-HA-CaSO<sub>4</sub> attenuated greatly during *in vitro* resorption (**Figure 7**). The CaSO<sub>4</sub> phase (1) was difficult to detect after Day 56 and HA (3) became the dominant inorganic phase detected. The organic phase, PCL (2), also remained within the 3DP scaffold structure, exhibiting diffraction peaks at  $2\theta = 21.3^{\circ}$ and  $23.7^{\circ}$ . This result indicated that CaSO<sub>4</sub> experienced a more intensive level of dissolution when compared to the HA phase. It is well known that CaSO<sub>4</sub> dissolves more rapidly than β-TCP and HA [34]. The loss of scaffold mass was attributed to the steady dissolution of CaSO<sub>4</sub>, 338 which was exposed directly to the buffered solution. The remaining CaSO<sub>4</sub> was encapsulated 339 by PCL and therefore, was unable to be detected towards the end of resorption. The absence of 340 CaSO<sub>4</sub> during the latter stages of resorption could provide several benefits: firstly, its rapid dissolution released a large amount of  $Ca^{2+}$  ions, which would favour cell proliferation *in vivo* 341 342 [35-36]. Secondly, the removal of the CaSO<sub>4</sub> left behind empty spaces that could enhance new 343 bone formation [37]. HA remained almost passive because the resorption of multiphase-344 ceramic always begins with the dissolution of the most soluble phase [37]. The greater dissolution of CaSO<sub>4</sub> created an environment rich in Ca<sup>2+</sup> ions, which also inhibited HA 345 dissolution. The SEM analysis showed that a large number of ceramic particles remained 346 347 within the 3DP scaffold at Day 56, which were connected by PCL fibrils (Figure 9). 348 Additionally, the scaffold surface was still well covered with the PCL coating layer. Therefore, 349 a high level of structural integrity was still available after the 3DP scaffold was subjected to 350 dynamic in vitro resorption for 56 days. This signifies the potential for the 3DP scaffold to 351 offer structural and mechanical support to the newly formed tissues.

352 During the 3D printing process, a solid structure was formed due to the precipitation of ceramic 353 aggregates when the setting reaction between the CaSO<sub>4</sub> powder and water-based binder 354 occurred (Figure 10). The aggregated rod-like crystals eroded when the structure was subjected 355 to resorption in the buffer solution. The surface erosion resulted in a smoother topography, 356 which was observed for the PCL-HA-CaSO<sub>4</sub> morphology after 28 days immersion in buffer 357 solution. More significant erosion was demonstrated when the resorption proceeded to Day 56. 358 These results demonstrated that the mechanism of surface erosion was dominant, whereby the 359 material located within the internal structure only resorbed following complete resorption of 360 the external surfaces.

361 One of the challenges of analysing resorption using  $\mu$ -CT is the continuous shift of the X-ray 362 absorption peaks on the grayscale histogram, which was proportional to the material density 363 [38]. A decrease in material density related to the microporosity created as a consequence of 364 the resorption process. A voxel containing an element of microporosity exhibited a relatively 365 low grayscale value. A more dramatic shift of the X-ray absorption peaks towards lower values 366 was observed on the grayscale histogram of PCL-CaSO<sub>4</sub> 3DP scaffolds compared to their PCL-367 HA-CaSO<sub>4</sub> counterparts (Figure 11). There was no distinct peak observed on the histogram 368 for PCL-CaSO<sub>4</sub> 3DP scaffolds between Day 42 and Day 56. To demonstrate material distribution at different stages of the resorption period, binary images of cross-sections of 3DP 369

370 scaffolds were generated via thresholding the reconstruction images at different grayscale 371 levels based on the histograms: (1) 24-255 for PCL-HA-CaSO<sub>4</sub> and (2) 5-37 and 37-255 for 372 PCL-CaSO<sub>4</sub> (Figure 12). The cross-sectional images of PCL-HA-CaSO<sub>4</sub> after 56 days of 373 resorption did not exhibit a significant change and a high level of structural integrity was 374 maintained. Smaller pores (approx. 100 µm) were created within the structure after 56 days of 375 resorption. However, PCL-CaSO<sub>4</sub> scaffolds experienced a rapid loss of material (Figure 13), 376 which correlated to a lower grayscale range at Day 28 as a consequence of the reduction in 377 density throughout the 3DP scaffold. This transformation continued to be more significant as 378 only a few materials were observed on the binary images at Day 56. The dip-coating technique 379 enabled deep infiltration of PCL within the scaffold structure, which efficiently filled the inter-380 particle spacing. The entire 3DP scaffold structure was maintained due to the PCL 381 demonstrating a low level of resorption by Day 56. Due to this infiltration of PCL within the 382 3DP scaffold it was highly unlikely that large spaces were created unless the *in vitro* resorption 383 reached a relatively high level, which was the case for the PCL-CaSO<sub>4</sub> 3DP scaffold type. 384 However, for the PCL-HA-CaSO<sub>4</sub> based 3DP scaffold, both PCL infiltration and faster 385 dissolution of CaSO<sub>4</sub> reduced HA dissolution. Consequently, a large amount of HA particles 386 remained within the 3DP scaffold structure.

387 This study has been focused on the printability of the polymer coated composite materials and 388 the corresponding *in vitro* data reported over a 56-day period shows promise. Notwithstanding 389 this fact, there may be additional questions that are still unresolved and areas for further 390 development. Although, use of the PCL coating successfully increased the compressive 391 strength to match cancellous bone (2-12 MPa) [47]. The compressive modulus, however, was 392 still lower than that of cancellous bone (100-500 MPa) due to the loose packing nature of 393 ceramic particles. Further improvement in the compressive properties could be achieved 394 through: (1) optimisation of the particle characteristics to achieve more dense powder packing, 395 or (2) incorporation of ceramic particles within the polymer infiltration solution for better 396 structural reinforcement. Further investigation is necessary to understand how the PCL-HA-397 CaSO4 based 3DP scaffolds perform under cyclic loading within a pseudo-physiological 398 environment as this is an essential next step before *in vivo* testing using an appropriate animal 399 model. Studies regarding the cytotoxic response of both HA and CaSO<sub>4</sub> based biomaterials 400 have documented favourable findings corroborating their potential in orthopaedic applications 401 [14,25,39-42,48]. However, there may be toxicity issues associated with our composite 402 scaffold since the materials were processed via a novel technology. It is important that 403 compressive and tunable resorption properties of the 3DP scaffolds are appropriately balanced 404 to the sequence of events exhibited during new bone formation in vivo, i.e. encouraging new 405 tissue growth during the initial stages of implantation, offering sufficient structural and 406 mechanical support and follow an appropriate resorption profile during the bone healing phase. 407 This balance can be achieved by varying the CaSO<sub>4</sub>:CaP:PCL of the final material formulation 408 to mimic the mechanical properties and resorption requirements for a specific clinical 409 application, which will vary considerably depending on number of factors, e.g. the age of 410 patient, bone quality, size and anatomical site of the defect. Further work is currently ongoing 411 to: (1) augment the mechanical properties of the existing 3DP scaffolds and understand how 412 they perform under cyclic loading within a pseudo-physiological environment, (2) determine 413 the extent of biocompatibility and osteogenic properties of these 3DP manufactured scaffolds 414 using an appropriate in vivo animal model and (3) understand if the 3DP scaffolds offer the 415 necessary mechanical and structural support and resorption rate required for complete repair of 416 a critical sized bone defect and coincide with sequence of events involved in natural bone 417 healing.

## 418 **4.** Conclusion

419 This study demonstrated that a blend of HA and CaSO<sub>4</sub> was a reliable formulation for powder-420 based 3DP technology. The combination of using a fast resorbing CaSO<sub>4</sub> powder and the 421 relatively slower HA powder represents a promising prospect for tuning the bioresorption of 422 3DP scaffolds. The HA-CaSO<sub>4</sub> could be tailored to coincide with tissue growth rate for different 423 surgical procedures. Following manufacture of the 3DP scaffold, PCL infiltration was used to 424 successfully fill the inter-particle spacing within the scaffold and as a result, the compressive 425 properties increased significantly. The *in vitro* resorption properties under a dynamic flow 426 conditions demonstrated PCL inhibited HA dissolution and maintained the overall structural 427 integrity after 56 days of resorption. The remaining scaffold structure accounted for 45.52  $\pm$ 428 0.54% of the original mass. Rapid reduction in the compressive properties during the first 28 429 days were mainly attributed to CaSO<sub>4</sub> dissolution. The majority of the CaSO<sub>4</sub> powder resorbed 430 within 56 days, while HA remained almost passive. The resorption of the CaSO<sub>4</sub> particles 431 allowed for spaces to become available, which could act as niches, encouraging new tissue 432 growth in the vicinity of the slow resorbing HA. The prepared HA-CaSO<sub>4</sub> scaffold with 433 enhanced mechanical properties and tuneable bioresorption has potential as a good candidate 434 for bone tissue engineering applications.

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Figure 1: (top) Experimental set-up for characterisation of *in vitro* resorption of 3DP scaffolds
under dynamic flow conditions. (bottom) Top and side views of the scaffold chamber.
Direction of fluid flow is indicated by the red arrows.



**Figure 2**: (a) Compressive strength, (b) compressive modulus, and (c) compressive toughness 588 (Mean  $\pm$  SD) for HA-CaSO<sub>4</sub> and PCL-HA-CaSO<sub>4</sub>. \*\* p < 0.05 and \*\*\* p < 0.001, indicating a

589 significant difference between HA-CaSO<sub>4</sub> and PCL-HA-CaSO<sub>4</sub>.



- **Figure 3**: Typical compressive deformation for HA-CaSO<sub>4</sub> and PCL-HA-CaSO<sub>4</sub> 3DP scaffolds
- and difference in scaffold integrity post-failure. Scale bar = 10 mm.



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**Figure 4**: SEM images of the surfaces of: (**a**) HA-CaSO<sub>4</sub>, arrows indicate inter-particle gaps (200x); (**b**) PCL-HA-CaSO<sub>4</sub> (190x); (**c**) PCL films filling inter-particle gaps, arrows indicate PCL coating infiltrating inter-particle gaps (900x); and (**d**) PCL-HA-CaSO<sub>4</sub> with PCL fibrils bridging the fractured surface, arrows indicate PCL bridging between ceramic particles within fractured surface (350x).



Figure 5: Mass change (Mean  $\pm$  SD) for PCL-HA-CaSO<sub>4</sub> and PCL-CaSO<sub>4</sub> following dynamic 

resorption up to 56 days. \*\*\* P-value<0.001, indicating a significant difference between PCL-HA-CaSO<sub>4</sub> and PCL-CaSO<sub>4</sub> at the same time point. 



**Figure 6**: Water absorption (Mean  $\pm$  SD) for PCL-HA-CaSO<sub>4</sub> and PCL-CaSO<sub>4</sub> following 611 dynamic resorption up to 56 days. \*\* p-value<0.001, indicating a significant difference 612 between PCL-HA-CaSO<sub>4</sub> and PCL-CaSO<sub>4</sub> at the same time point.



**Figure 7**: XRD spectra for PCL-HA-CaSO<sub>4</sub> after Day 7, 28, and 56 of dynamic *in vitro* 616 resorption.



620 Figure 8: (a) Compressive strength and (b) compressive modulus (Mean  $\pm$  SD) for PCL-CaSO<sub>4</sub>

and PCL-HA-CaSO<sub>4</sub> following resorption up to 56 days. The average percentage change of the
 properties at each time point is also shown as a line plot referring to the labelled secondary Y-

623 axis.



- **Figure 9**: SEM images of (**a**) the internal structure and (**b**) the surface of PCL-HA-CaSO<sub>4</sub> after
- 627 Day 56 of dynamic resorption, arrows indicate PCL fibrils.



Figure 10: SEM images of the ceramic surfaces on the internal structure of PCL-HA-CaSO<sub>4</sub>
after Day 0, 28, and 56 of dynamic resorption.



**Figure 11**: Grayscale histograms for μ-CT results of: (**a**) PCL-HA-CaSO<sub>4</sub> and (**b**) PCL-CaSO<sub>4</sub>

637 following dynamic resorption up to Day 56.



Figure 12: Reconstruction and binary images of the cross sections of PCL-HA-CaSO<sub>4</sub> and PCL-CaSO<sub>4</sub> at four time-points of dynamic degradation (Day 0, 7, 28 and 56). The binary images was obtained using thresholding at two levels (37-255 and 5-37) for PCL-CaSO<sub>4</sub> and at one level (24-255) for PCL-HA-CaSO<sub>4</sub>. Pixels having grayscale in the range denoted in black. Scale bar = 2 mm.





647 Figure 13: Results determined by  $\mu$ -CT analysis for material volume loss of PCL-HA-CaSO<sub>4</sub>

and PCL-CaSO<sub>4</sub> scaffolds as a function of dynamic resorption up to Day 56.

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