Total Synthesis of the GRP78-Downregulatory Macrolide (+)-Prunustatin A, the Immunosuppressant (+)-SW-163A, and a JBIR-04 Diastereoisomer That Confirms JBIR-04 Has Nonidentical Stereochemistry to (+)-Prunustatin A

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Supporting Information

ABSTRACT: A unified total synthesis of the GRP78-downregulator (+)-prunustatin A and the immunosuppressant (+)-SW-163A based upon [1 + 1 + 1 + 1]-fragment condensation and macrolactonization between O(4) and C(5) is herein described. Sharpless asymmetric dihydroxylation was used to set the C(2) stereocenter present in both targets. In like fashion, coupling of the (+)-prunustatin A macrolide amine with benzoic acid furnished a JBIR-04 diastereoisomer whose NMR spectra did not match those of JBIR-04, thus confirming that it has different stereochemistry than (+)-prunustatin A.

This has prompted a number of groups to devise elegant total syntheses of (+)-prunustatin A to increase the supply, with the teams of Kawanishi and Usuki scoring particularly notable successes in this regard.

From a therapeutic perspective, (+)-prunustatin A is of considerable pharmaceutical interest because of its pronounced downregulatory effects on GRP78/BIP (78 kDa glucose-regulated protein) expression in glucose-deprived HT1080 human fibrosarcoma cells at very low drug concentrations (IC_{50} = 11.5 nM), with total inhibition of GRP78 expression occurring at the 80 nM level and full cancer cell apoptosis occurring at the slightly higher drug concentration of 100 nM. Importantly, (+)-prunustatin A is non-cytotoxic toward HT1080 cells under normal conditions, where it functions as a cytostatic agent even at concentrations as high as 500 nM. This remarkable property of (+)-prunustatin A to selectively induce apoptosis within highly stressed, glucose-deprived, cancer cells suggests that it might potentially be useful to combat hypoxic human tumors while leaving normal healthy tissue undamaged.

Upregulated GRP78 expression within hypoxic solid tumors is now thought to contribute significantly toward them becoming refractory toward treatment with drugs and radiotherapy. There is thus a very good medical case for clinically establishing whether (+)-prunustatin A will be of value for treating such cancers. However, preliminary screening of (+)-prunustatin A against drug-resistant tumors, Gram-negative Neisseria gonorrhoeae and Neisseria meningitides strains, and they might also prove useful for counteracting many lethal viral infections. In the latter regard, many viruses rely on GRP78-regulated machinery to create functionally active virions (e.g., the Ebola, Lassa, and Marburg hemorrhagic RNA viruses). Because of this, we became interested in developing a new synthetic route to (+)-prunustatin A to expedite its future clinical development and that of its powerful reduced immunosuppressant congener, (+)-SW-163A. In this Letter we report our success in these endeavors.

Following several abortive attempts to synthesize (+)-prunustatin A by strategies involving macrocyclization between O(7) and C(8), w e report our success in these endeavors.

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desilylation, with tetraester 4 being assembled from the sterically hindered acid 9 by two successive esterifications, the first involving 8 and 9 and the second involving 5 and 6. Keto acid 5 would be derived from triester 7 by O-debenzylation at C(1), oxidation of the alcohol to the ketone, and O-deallylation. Acid 9 was envisioned to emanate from alcohol 12 by esterification with protected L-lactic acid derivative 11 allied with selective O-desilylation and oxidation of the primary alcohol to the acid. The main issue associated with constructing alcohol 12 would be correct positioning of the various protecting groups to enable the alcohol at C(2) to be selectively presented to 11. For this, we hoped to take advantage of a Sharpless asymmetric dihydroxylation (AD) reaction on (E)-alkene 14 accompanied by in situ lactonization to give the corresponding β-hydroxybutyrolactone with high ee. If successful, such an approach would nicely allow protection of the secondary hydroxyl at C(1) as an O-p-methoxybenzyl (OPMB) ether, as in compound 13, to allow the remaining features to be elaborated by reduction and selective O-desilylation. Of course, a good option for constructing the (E)-olefin in 14 would be a Julia–Kocienski olefination between tetrazolylsulfone 15 and known β-aldehydo ester 16.

Our new (+)-prunustatin A campaign began with a repetition of the synthesis of β-aldehydo ester 16, which was typically prepared in 67% yield from 17 by pyridinium chlorochromate (PCC) oxidation in CH2Cl2 (Scheme 2). Aldehyde 16 was then condensed with the anion derived from 15 to give (E)-alkene 14 in 64% yield with total stereocontrol. While initially we accessed 19 in near optically pure condition via Sharpless AD with AD-mix-β,7a we found this process to be inconveniently slow, needing 7 days to reach completion. We therefore evaluated less hindered Sharpless ligands for this purpose. A considerable improvement in the reaction rate was found when the AD was conducted with catalytic potassium osmate (1 mol %) and the DHQD-MEQ ligand7b (13 mol %), which afforded the lactonized product 19 in 95% yield with 100% ee after only 30 h of stirring at 0 °C, which represented a considerable
operational improvement and was also much cheaper to carry out.

The alcohol in 19 was then protected as an OPMB ether with NaH and PMBCl in DMF, and the product, lactone 13, was reduced with LiAlH4 to obtain diol 20 in 85% yield. The less hindered primary hydroxyl of 20 was next regioselectively protected as an O-trimethylsilyl (OTES) ether to allow the all-important ester bond to be grafted onto O(2). For this, (S)-lactic acid derivative 21 was first converted into the Yamaguchi10, 24,6-trichlorobenzoic acid mixed anhydride 22, and this was reacted with 12 in CH2Cl2 at rt for 17 h in the presence of 4-(dimethylamino)pyridine (DMAP) (3.25 equiv). This proved to be the optimal method for esterifying this system, furnishing 10 in 93% yield. Having reliably fulfilled its alcohol-differentiating role, the primary OTES ether was selectively cleaved from 10 by catalytic pyridinium p-toluenesulfonate (PPTS) (0.1 equiv) in MeOH over 45 min at rt. A two-stage oxidation thereafter converted alcohol 23 into carboxylic acid 9. In this sequence, a Ley—Griffith catalytic n-Pr4NRuO4/N-methylmorpholine N-oxide (NMO) oxidation11 first furnished aldehyde 24 in near quantitative yield, and a Pinnick oxidation subsequently provided 9 in 89% yield.12

We next focused our attention on converting L-isoleucine (25) into L-isoleucic acid (26) and the latter into O-allyl ester 8 (Scheme 3).13 To access the former, we followed the diazotation procedure of Plenkiewicz and Poterala,13a which worked very well in our hands, and generated the HNO2 in situ from 1 M aqueous H2SO4 and NaNO2. It delivered the crystalline K2CO3/allyl bromide/Bu4NI in DMF at rt; the product ester was isolated in 82% yield. It was then coupled to acid 9 using excess N-(3-(dimethylamino)propyl)-N-ethyl carbodiimide hydrochloride (EDCI) and DMAP as the acid activators; triester 7 was formed in 72% yield after 18 h of stirring at rt in CH2Cl2.

DDQ was now used to chemoselectively remove the PMB group from O(1) without disturbing the potentially sensitive O-allyl ester. The resulting alcohol 27 was oxidized to the ketone with n-Pr4NRuO4/NMO in MeCN, and the O-allyl ester was detached with PhSiH3 and Pd(0). The acid 5 so produced was then coupled to partially protected L-threonine derivative 6 using 2-methyl-6-nitrobenzoic anhydride (MNBA)14 and DMAP in CH2Cl2, affording the desired product 4 in 74% yield. Subsequently, 4 was O-deallylated with PhSiH3 and catalytic Pd(PPh3)4 in CH2Cl2,15 and the product acid was O-desilylated with HF-pyridine complex in a mixture of pyridine/THF. Both deprotections proceeded cleanly to provide the required seco-acid 3 in 82% yield over the two steps. The latter was then macrolactonized on a 0.3 g scale under high-dilution conditions by addition of a solution of 3 in CH2Cl2/THF (1:1) over 9 h to a solution of DMAP (2 equiv) and MNBA14 (1.3 equiv) in dry THF at rt, attaining a final reaction concentration of ca. 0.00048 M with respect to 3. The reactants were then allowed to stir at rt for 39 h to bring about the desired ring closure. Macrolide 29 was isolated pure in 42% yield after SiO2 flash chromatography. The structure of 29 was unambiguously confirmed by single-crystal X-ray analysis (see the Supporting Information). Importantly, the 400 MHz 1H NMR spectrum of 29 in CDCl3 matched that of Kawanishi.3

Although we did attempt to repeat the 50 °C macro lactonization protocol of Yamakoshi and Kawanishi5,16 on 3 at the reaction concentration of 0.0012 M that they reported, we found it extremely difficult to control the rate of the addition of the seco-acid solution to the MNBA/DMAP solution, when either a syringe pump or slow cannulation was used to deliver the THF/CH2Cl2 solution of 3. Not only did the hot vapor from the reaction mixture consistently oppose a carefully controlled slow addition of the solution of 3 into the reaction flask, but also, the heating process caused much more variable reaction outcomes, with the attendant formation of more complex mixtures. Our very best yield of 29 from adhering to the 50 °C cyclization
protocol in ref 3 was 49%, but this was not the norm. Because of the significant technical difficulties and reaction variations that attend this method, we recommend that other workers use the much more consistent ultrahigh-dilution rt cyclization procedure that we have described in the Supporting Information. However, even under our rt conditions, intermolecular dimerization of 3 still continues to be significant, but generally less so than under the 50 °C reaction conditions.3,16

In order to complete our synthesis of (+)-prunustatin A, the Boc group of 29 was detached with neat CF3CO2H in CH2Cl2, and the crude TFA salt 30 was coupled with 1,3,4,18 using EDCI, N-ethylmorpholine (NEM), and 1-hydroxybenzotriazole (HOBT). The desired product 31 was isolated in 63% yield after SiO2 flash chromatography; it was identical to the same compound prepared by Usuki.1 Compound 31 was then deprotected by catalytic hydrogenation with 10% Pd/C in EtOAc/Methanol (1:1) at 1 atm; synthetic (+)-prunustatin A was isolated in 65% yield after SiO2 chromatography (0.74% overall). Its spectroscopic values closely matched those reported by Shin-ya,13 Kawanishi,1 and Usuki,1 thus confirming that the natural product had indeed been synthesized. NaBH4 reduction of (+)-prunustatin A in EtOH also furnished the immunosuppressive (-)-SW-163A,6 in accord with Shin-ya,1,2 Kawanishi,3 and Usuki,4 thus confirming the absolute stereochemistry of JBIR-04 diastereomer 32.18 The latter synthesis also revealed that the absolute stereochemistry of JBIR-04 differs from that found in (+)-prunustatin A. We expect that our new synthetic pathway to JBIR-04 will continue to be signifi-
cant, but generally less so than under the 50 °C reaction conditions, intermolecular dimerization of 3 still continues to be significant, but generally less so than under the 50 °C reaction conditions.3,16

Given that we had unambiguously proven the stereochemistry of 29, we next deprotected its Boc group and coupled 30 to PhCO2H in order to secure what we hoped was going to be the structurally related natural product JBIR-0417 (Scheme 4), whose absolute stereostructure has not been assigned to date. Unfortunately, our spectroscopic comparisons of 32 with JBIR-
04 soon confirmed that JBIR-04 has different absolute stereochemistry than (+)-prunustatin A, which perhaps explains why its GRP78-downregulatory effects are 200 times lower. In conclusion, we have devised a highly stereoselective total syntheses of (+)-prunustatin A, SW-163A, and JBIR-04 diastereoisomer 32.18 The latter synthesis also revealed that the absolute stereochemistry of JBIR-04 differs from that found in (+)-prunustatin A. We expect that our new synthetic pathway to these molecules will prove useful for fashioning analogues, including biotinylated ones, which would have potential value for new drug target retrieval by affinity chromatography.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b01235.

Full experimental procedures for all steps, copies of the IR, HRMS, and 1H,13C NMR spectra of every intermediate, and X-ray plots and crystallographic data for 19, 13, and 29 (including CCDC accession numbers) (PDF)

SQUEEZE-processed crystallographic data for 29 (CIF)

Original crystallographic data for 29 (CIF)

Crystallographic data for 13 (CIF)

Crystallographic data for 19 (CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

(16) Because ref 3 reports no experimental procedures at all, we reproduced as best we could the 50 °C macrocyclization in THF/CH2Cl2 outlined in that paper.